

# 1(2H)-Naphthalenone, 3,4-dihydro-4-methyl-

<b>Other names:</b>	4-Methyl-«alpha»-tetralone 4-Methyl-1-tetralone 3,4-Dihydro-4-methyl-1(2H)-naphthalenone 1,2,3,4-tetrahydro-4-methylnaphthalen-1-one
<b>Inchi:</b>	InChI=1S/C11H12O/c1-8-6-7-11(12)10-5-3-2-4-9(8)10/h2-5,8H,6-7H2,1H3
<b>InchiKey:</b>	SRLHDEROUKFEMJ-UHFFFAOYSA-N
<b>Formula:</b>	C11H12O
<b>SMILES:</b>	CC1CCC(=O)c2cccc21
<b>Mol. weight [g/mol]:</b>	160.21
<b>CAS:</b>	19832-98-5

## Physical Properties

Property code	Value	Unit	Source
gf	70.58	kJ/mol	Joback Method
hf	-116.37	kJ/mol	Joback Method
hfus	13.44	kJ/mol	Joback Method
hvap	47.35	kJ/mol	Joback Method
log10ws	-3.21		Crippen Method
logp	2.767		Crippen Method
mcvol	132.800	ml/mol	McGowan Method
pc	3199.16	kPa	Joback Method
tb	561.57	K	Joback Method
tc	805.91	K	Joback Method
tf	335.31	K	Joback Method
vc	0.499	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	316.27	J/molxK	561.57	Joback Method
cpg	333.21	J/molxK	602.29	Joback Method
cpg	349.06	J/molxK	643.02	Joback Method
cpg	363.84	J/molxK	683.74	Joback Method
cpg	377.60	J/molxK	724.46	Joback Method

cpg	390.35	J/mol×K	765.19	Joback Method
cpg	402.13	J/mol×K	805.91	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	368.20	K	0.10	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19832985&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19832985&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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