

# Abieta-7,13-dien-3-one

<b>Other names:</b>	Abieta-7,13-diene-3-one (4aR,4bR,10aR)-7-Isopropyl-1,1,4a-trimethyl-3,4,4a,4b,5,6,10,10a-octahydrophenanthrene
<b>Inchi:</b>	InChI=1S/C20H30O/c1-13(2)14-6-8-16-15(12-14)7-9-17-19(3,4)18(21)10-11-20(16,17)5/
<b>InchiKey:</b>	BOPYLVIBVAFZAH-HLIPFELVSA-N
<b>Formula:</b>	C20H30O
<b>SMILES:</b>	<chem>CC(C)C1=CC2=CCC3C(C)(C)C(=O)CCC3(C)C2CC1</chem>
<b>Mol. weight [g/mol]:</b>	286.45
<b>CAS:</b>	29461-25-4

## Physical Properties

Property code	Value	Unit	Source
gf	136.21	kJ/mol	Joback Method
hf	-308.75	kJ/mol	Joback Method
hfus	17.59	kJ/mol	Joback Method
hvap	63.87	kJ/mol	Joback Method
log10ws	-5.66		Crippen Method
logp	5.321		Crippen Method
mcvol	253.050	ml/mol	McGowan Method
pc	1624.60	kPa	Joback Method
rinpol	2313.00		NIST Webbook
rinpol	2302.00		NIST Webbook
rinpol	2313.00		NIST Webbook
rinpol	2297.00		NIST Webbook
rinpol	2302.00		NIST Webbook
rinpol	2302.00		NIST Webbook
rinpol	2302.00		NIST Webbook
rinpol	2302.00		NIST Webbook
rinpol	2297.00		NIST Webbook
rinpol	2315.00		NIST Webbook
rinpol	2303.00		NIST Webbook
rinpol	2335.50		NIST Webbook
tb	770.04	K	Joback Method
tc	1013.89	K	Joback Method
tf	474.72	K	Joback Method
vc	0.955	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	810.44	J/mol×K	770.04	Joback Method
cpg	835.91	J/mol×K	810.68	Joback Method
cpg	860.76	J/mol×K	851.32	Joback Method
cpg	885.32	J/mol×K	891.96	Joback Method
cpg	909.90	J/mol×K	932.60	Joback Method
cpg	934.82	J/mol×K	973.24	Joback Method
cpg	960.39	J/mol×K	1013.89	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C29461254&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C29461254&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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