

# Dimyrcene I-a

<b>Inchi:</b>	InChI=1S/C20H34/c1-6-20(14-8-10-18(4)5)15-12-19(13-16-20)11-7-9-17(2)3/h6,9-10,19H
<b>InchiKey:</b>	ILNRJDGFLRCSOE-UHFFFAOYSA-N
<b>Formula:</b>	C20H34
<b>SMILES:</b>	<chem>C=CC1(CCC=C(C)C)CCC(CCC=C(C)C)CC1</chem>
<b>Mol. weight [g/mol]:</b>	274.48

## Physical Properties

Property code	Value	Unit	Source
gf	359.95	kJ/mol	Joback Method
hf	-66.62	kJ/mol	Joback Method
hfus	30.67	kJ/mol	Joback Method
hvap	58.49	kJ/mol	Joback Method
log10ws	-7.17		Crippen Method
logp	6.842		Crippen Method
mvol	268.900	ml/mol	McGowan Method
pc	1308.01	kPa	Joback Method
ripol	2144.00		NIST Webbook
ripol	2149.00		NIST Webbook
tb	676.88	K	Joback Method
tc	881.24	K	Joback Method
tf	302.36	K	Joback Method
vc	1.028	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	768.88	J/mol×K	676.88	Joback Method
cpg	792.30	J/mol×K	710.94	Joback Method
cpg	814.58	J/mol×K	745.00	Joback Method
cpg	835.86	J/mol×K	779.06	Joback Method
cpg	856.31	J/mol×K	813.12	Joback Method
cpg	876.05	J/mol×K	847.18	Joback Method
cpg	895.25	J/mol×K	881.24	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R407175&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R407175&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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