

2,6-Difluoro-3-methylbenzoic acid, eicosyl ester

Inchi: InChI=1S/C28H46F2O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-23-32-28(3)
InchiKey: QEFDIZQXWJOHMF-UHFFFAOYSA-N
Formula: C28H46F2O2
SMILES: CCCCCCCCCCCCCCCCCCCCCOC(=O)c1c(F)ccc(C)c1F
Mol. weight [g/mol]: 452.66

Physical Properties

Property code	Value	Unit	Source
gf	-355.14	kJ/mol	Joback Method
hf	-1056.15	kJ/mol	Joback Method
hfus	70.10	kJ/mol	Joback Method
hvap	89.71	kJ/mol	Joback Method
log10ws	-10.81		Crippen Method
logp	9.472		Crippen Method
mcvol	392.600	ml/mol	McGowan Method
pc	744.88	kPa	Joback Method
rinpol	3186.00		NIST Webbook
tb	956.49	K	Joback Method
tc	1175.43	K	Joback Method
tf	542.64	K	Joback Method
vc	1.556	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1333.55	J/mol×K	956.49	Joback Method
cpg	1354.11	J/mol×K	992.98	Joback Method
cpg	1373.14	J/mol×K	1029.47	Joback Method
cpg	1390.70	J/mol×K	1065.96	Joback Method
cpg	1406.85	J/mol×K	1102.45	Joback Method
cpg	1421.65	J/mol×K	1138.94	Joback Method
cpg	1435.18	J/mol×K	1175.43	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U338857&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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