

# Ethylvinyl sulfide

<b>Other names:</b>	CH <sub>2</sub> =CHSC <sub>2</sub> H <sub>5</sub> Ethene, (ethylthio)- Sulfide, ethyl vinyl Vinyl ethyl thioether (Ethylthio)ethene
<b>Inchi:</b>	InChI=1S/C4H8S/c1-3-5-4-2/h3H,1,4H2,2H3
<b>InchiKey:</b>	AFGACPRTZOCNIW-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>4</sub> H <sub>8</sub> S
<b>SMILES:</b>	C=CSCC
<b>Mol. weight [g/mol]:</b>	88.17
<b>CAS:</b>	627-50-9

## Physical Properties

Property code	Value	Unit	Source
gf	103.76	kJ/mol	Joback Method
hf	41.41	kJ/mol	Joback Method
hfus	8.97	kJ/mol	Joback Method
hvap	30.65	kJ/mol	Joback Method
ie	8.21 ± 0.01	eV	NIST Webbook
log10ws	-1.73		Crippen Method
logp	1.883		Crippen Method
mcvol	79.270	ml/mol	McGowan Method
pc	4200.18	kPa	Joback Method
rinpol	696.00		NIST Webbook
rinpol	696.00		NIST Webbook
rinpol	696.00		NIST Webbook
tb	366.50 ± 1.50	K	NIST Webbook
tc	551.75	K	Joback Method
tf	167.48	K	Joback Method
vc	0.294	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	122.29	J/mol×K	356.38	Joback Method
cpg	130.03	J/mol×K	388.94	Joback Method
cpg	137.45	J/mol×K	421.50	Joback Method
cpg	144.57	J/mol×K	454.07	Joback Method
cpg	151.40	J/mol×K	486.63	Joback Method
cpg	157.93	J/mol×K	519.19	Joback Method
cpg	164.17	J/mol×K	551.75	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C627509&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C627509&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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