

# 2-Butanamine, N-methyl-

<b>Other names:</b>	sec-Butylmethylamine Propylamine, N,1-dimethyl- N-Methyl-sec-butylamine N-Methyl-N-sec-butylamine N-Methyl-2-butylamine N,1-Dimethylpropylamine 2-(Methylamino)butane N-(sec-Butyl)-N-methylamine
<b>Inchi:</b>	InChI=1S/C5H13N/c1-4-5(2)6-3/h5-6H,4H2,1-3H3
<b>InchiKey:</b>	PYFSCIWXNSXGNS-UHFFFAOYSA-N
<b>Formula:</b>	C5H13N
<b>SMILES:</b>	CCC(C)NC
<b>Mol. weight [g/mol]:</b>	87.16
<b>CAS:</b>	7713-69-1

## Physical Properties

Property code	Value	Unit	Source
gf	78.17	kJ/mol	Joback Method
hf	-98.34	kJ/mol	Joback Method
hfus	10.28	kJ/mol	Joback Method
hvap	32.77	kJ/mol	Joback Method
log10ws	-1.22		Crippen Method
logp	1.004		Crippen Method
mcvol	91.290	ml/mol	McGowan Method
pc	3555.77	kPa	Joback Method
rinpol	652.00		NIST Webbook
tb	363.53	K	Joback Method
tc	537.86	K	Joback Method
tf	183.77	K	Joback Method
vc	0.344	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	162.53	J/mol×K	363.53	Joback Method
cpg	172.97	J/mol×K	392.59	Joback Method
cpg	183.01	J/mol×K	421.64	Joback Method
cpg	192.67	J/mol×K	450.70	Joback Method
cpg	201.96	J/mol×K	479.75	Joback Method
cpg	210.89	J/mol×K	508.81	Joback Method
cpg	219.45	J/mol×K	537.86	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7713691&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7713691&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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