

Pentane, 3-iodo-3-methyl

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|----------------------|---|
| Inchi: | InChI=1S/C6H13I/c1-4-6(3,7)5-2/h4-5H2,1-3H3 |
| InchiKey: | CIDZNASUPYMQCG-UHFFFAOYSA-N |
| Formula: | C6H13I |
| SMILES: | CCC(C)(I)CC |
| Mol. weight [g/mol]: | 212.07 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|---------|----------------|
| gf | 60.60 | kJ/mol | Joback Method |
| hf | -99.05 | kJ/mol | Joback Method |
| hfus | 8.29 | kJ/mol | Joback Method |
| hvap | 37.03 | kJ/mol | Joback Method |
| log10ws | -3.39 | | Crippen Method |
| logp | 3.000 | | Crippen Method |
| mcvol | 121.220 | ml/mol | McGowan Method |
| pc | 3096.73 | kPa | Joback Method |
| rinpol | 958.00 | | NIST Webbook |
| rinpol | 1023.00 | | NIST Webbook |
| ripol | 1179.00 | | NIST Webbook |
| tb | 426.59 | K | Joback Method |
| tc | 640.95 | K | Joback Method |
| tf | 217.86 | K | Joback Method |
| vc | 0.449 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 211.97 | J/molxK | 426.59 | Joback Method |
| cpg | 266.25 | J/molxK | 605.22 | Joback Method |
| cpg | 256.82 | J/molxK | 569.49 | Joback Method |
| cpg | 246.73 | J/molxK | 533.77 | Joback Method |
| cpg | 235.92 | J/molxK | 498.04 | Joback Method |
| cpg | 224.35 | J/molxK | 462.32 | Joback Method |
| cpg | 275.05 | J/molxK | 640.95 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0003828 | Paxs | 426.59 | Joback Method |
| dvisc | 0.0005200 | Paxs | 391.80 | Joback Method |
| dvisc | 0.0007498 | Paxs | 357.01 | Joback Method |
| dvisc | 0.0011699 | Paxs | 322.23 | Joback Method |
| dvisc | 0.0020332 | Paxs | 287.44 | Joback Method |
| dvisc | 0.0041142 | Paxs | 252.65 | Joback Method |
| dvisc | 0.0104269 | Paxs | 217.86 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=R25594&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| ripol: | Polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/77-807-1/Pentane-3-iodo-3-methyl.pdf>

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