

# 1,1,1,2,2,3,3-Heptachloropropane

<b>Other names:</b>	ASYM-HEPTACHLOROPROPANE Propane, 1,1,1,2,2,3,3-heptachloro-
<b>Inchi:</b>	InChI=1S/C3HCl7/c4-1(5)2(6,7)3(8,9)10/h1H
<b>InchiKey:</b>	YFIENAGGCUHIQ-UHFFFAOYSA-N
<b>Formula:</b>	C3HCl7
<b>SMILES:</b>	<chem>C1C(Cl)C(Cl)(Cl)C(Cl)(Cl)Cl</chem>
<b>Mol. weight [g/mol]:</b>	285.21
<b>CAS:</b>	594-89-8

## Physical Properties

Property code	Value	Unit	Source
gf	-105.89	kJ/mol	Joback Method
hf	-238.21	kJ/mol	Joback Method
hfus	14.55	kJ/mol	Joback Method
hvap	49.99	kJ/mol	Joback Method
log10ws	-4.47		Crippen Method
logp	4.334		Crippen Method
mvol	138.810	ml/mol	McGowan Method
pc	3384.14	kPa	Joback Method
tb	520.70	K	NIST Webbook
tc	777.32	K	Joback Method
tf	322.85	K	Joback Method
vc	0.518	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	211.37	J/molxK	523.15	Joback Method
cpg	216.21	J/molxK	565.51	Joback Method
cpg	220.21	J/molxK	607.87	Joback Method
cpg	223.45	J/molxK	650.24	Joback Method
cpg	226.06	J/molxK	692.60	Joback Method
cpg	228.13	J/molxK	734.96	Joback Method
cpg	229.76	J/molxK	777.32	Joback Method

dvisc	0.0050546	Paxs	322.85	Joback Method
dvisc	0.0025844	Paxs	356.23	Joback Method
dvisc	0.0014823	Paxs	389.62	Joback Method
dvisc	0.0009282	Paxs	423.00	Joback Method
dvisc	0.0006224	Paxs	456.38	Joback Method
dvisc	0.0004407	Paxs	489.77	Joback Method
dvisc	0.0003261	Paxs	523.15	Joback Method
hvapt	34.80	kJ/mol	443.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	438.70	K	12.00	NIST Webbook
tbrp	405.20	K	4.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C594898&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C594898&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.cheric.org/files/research/kdb/mol/mol1582.mol">https://www.cheric.org/files/research/kdb/mol/mol1582.mol</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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