

# Ethanol, 1-(1-cyclohexenyl)-

<b>Other names:</b>	2-(1-cyclohexenyl)ethanol
<b>Inchi:</b>	InChI=1S/C8H14O/c1-7(9)8-5-3-2-4-6-8/h5,7,9H,2-4,6H2,1H3
<b>InchiKey:</b>	FAYUSHUTMRYHAD-UHFFFAOYSA-N
<b>Formula:</b>	C8H14O
<b>SMILES:</b>	CC(O)C1=CCCCC1
<b>Mol. weight [g/mol]:</b>	126.20
<b>CAS:</b>	3197-68-0

## Physical Properties

Property code	Value	Unit	Source
gf	-70.29	kJ/mol	Joback Method
hf	-244.99	kJ/mol	Joback Method
hfus	8.64	kJ/mol	Joback Method
hvap	51.38	kJ/mol	Joback Method
log10ws	-2.29		Crippen Method
logp	1.868		Crippen Method
mcvol	114.290	ml/mol	McGowan Method
pc	3777.68	kPa	Joback Method
tb	502.54	K	Joback Method
tc	700.57	K	Joback Method
tf	250.64	K	Joback Method
vc	0.416	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	257.62	J/mol×K	502.54	Joback Method
cpg	271.00	J/mol×K	535.55	Joback Method
cpg	283.67	J/mol×K	568.55	Joback Method
cpg	295.66	J/mol×K	601.56	Joback Method
cpg	306.99	J/mol×K	634.56	Joback Method
cpg	317.68	J/mol×K	667.57	Joback Method
cpg	327.77	J/mol×K	700.57	Joback Method
dvisc	0.0619163	Paxs	250.64	Joback Method

dvisc	0.0108281	Paxs	292.62	Joback Method
dvisc	0.0029331	Paxs	334.61	Joback Method
dvisc	0.0010631	Paxs	376.59	Joback Method
dvisc	0.0004723	Paxs	418.57	Joback Method
dvisc	0.0002433	Paxs	460.56	Joback Method
dvisc	0.0001400	Paxs	502.54	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3197680&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3197680&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/77-967-4/Ethanol-1-1-cyclohexenyl.pdf>

Generated by Cheméo on 2024-04-27 17:04:26.07399218 +0000 UTC m=+16526714.994569491.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.