

# 1-Aminoindan

<b>Other names:</b>	1-Aminoindane 1H-Inden-1-amine, 2,3-dihydro- 1-Indanamine indan-1-ylamine
<b>Inchi:</b>	InChI=1S/C9H11N/c10-9-6-5-7-3-1-2-4-8(7)9/h1-4,9H,5-6,10H2
<b>InchiKey:</b>	XJEVHMGJSYVQBQ-UHFFFAOYSA-N
<b>Formula:</b>	C9H11N
<b>SMILES:</b>	NC1CCc2ccccc21
<b>Mol. weight [g/mol]:</b>	133.19
<b>CAS:</b>	34698-41-4

## Physical Properties

Property code	Value	Unit	Source
gf	254.88	kJ/mol	Joback Method
hf	102.56	kJ/mol	Joback Method
hfus	16.05	kJ/mol	Joback Method
hvap	49.12	kJ/mol	Joback Method
log10ws	-2.45		Crippen Method
logp	1.633		Crippen Method
mcvol	113.030	ml/mol	McGowan Method
pc	4010.84	kPa	Joback Method
rinpol	1188.60		NIST Webbook
rinpol	207.63		NIST Webbook
rinpol	207.63		NIST Webbook
rinpol	1188.60		NIST Webbook
tb	516.25	K	Joback Method
tc	754.89	K	Joback Method
tf	331.33	K	Joback Method
vc	0.417	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	256.72	J/molxK	516.25	Joback Method

cpg	271.36	J/mol×K	556.02	Joback Method
cpg	284.89	J/mol×K	595.80	Joback Method
cpg	297.39	J/mol×K	635.57	Joback Method
cpg	308.95	J/mol×K	675.34	Joback Method
cpg	319.64	J/mol×K	715.12	Joback Method
cpg	329.54	J/mol×K	754.89	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	369.70	K	1.00	NIST Webbook
tbrp	493.60	K	99.60	NIST Webbook
tbrp	369.50 ± 0.50	K	1.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C34698414&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C34698414&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcpvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tbrp:** Boiling point at reduced pressure  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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