

# Aletamine

<b>Other names:</b>	«alpha»-Allylphenethylamine Alfetamine Benzeneethanamine, «alpha»-2-propenyl- «alpha»-2-Propenyl benzeneethanamine
<b>Inchi:</b>	InChI=1S/C11H15N/c1-2-6-11(12)9-10-7-4-3-5-8-10/h2-5,7-8,11H,1,6,9,12H2
<b>InchiKey:</b>	WQKXQJYCZMWOSD-UHFFFAOYSA-N
<b>Formula:</b>	C11H15N
<b>SMILES:</b>	<chem>C=CCC(N)Cc1ccccc1</chem>
<b>Mol. weight [g/mol]:</b>	161.24
<b>CAS:</b>	4255-23-6

## Physical Properties

Property code	Value	Unit	Source
gf	306.00	kJ/mol	Joback Method
hf	120.10	kJ/mol	Joback Method
hfus	18.68	kJ/mol	Joback Method
hvap	51.94	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	2.133		Crippen Method
mcvol	147.770	ml/mol	McGowan Method
pc	2966.57	kPa	Joback Method
rinpol	1293.00		NIST Webbook
rinpol	1311.00		NIST Webbook
rinpol	1293.00		NIST Webbook
rinpol	1320.00		NIST Webbook
rinpol	1311.00		NIST Webbook
ripol	1814.00		NIST Webbook
ripol	1814.00		NIST Webbook
tb	546.53	K	Joback Method
tc	768.13	K	Joback Method
tf	306.65	K	Joback Method
vc	0.547	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	343.55	J/molxK	546.53	Joback Method
cpg	359.08	J/molxK	583.46	Joback Method
cpg	373.60	J/molxK	620.40	Joback Method
cpg	387.15	J/molxK	657.33	Joback Method
cpg	399.78	J/molxK	694.26	Joback Method
cpg	411.56	J/molxK	731.20	Joback Method
cpg	422.54	J/molxK	768.13	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4255236&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4255236&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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