

# 1-Triacontanol

<b>Other names:</b>	Melissyl alcohol Myricyl alcohol Triacontanol-1 Triacontyl alcohol n-Triacontanol triacontan-1-ol
<b>Inchi:</b>	InChI=1S/C30H62O/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21-22-23-24
<b>InchiKey:</b>	REZQBEBOWJAQKS-UHFFFAOYSA-N
<b>Formula:</b>	C30H62O
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCO
<b>Mol. weight [g/mol]:</b>	438.81
<b>CAS:</b>	593-50-0

## Physical Properties

Property code	Value	Unit	Source
gf	64.90	kJ/mol	Joback Method
hf	-814.76	kJ/mol	Joback Method
hfus	77.54	kJ/mol	Joback Method
hvap	99.05	kJ/mol	Joback Method
log10ws	-11.64		Crippen Method
logp	10.921		Crippen Method
mvol	439.430	ml/mol	McGowan Method
pc	623.13	kPa	Joback Method
rinpol	3306.00		NIST Webbook
rinpol	3306.00		NIST Webbook
tb	977.98	K	Joback Method
tc	1233.03	K	Joback Method
tf	488.68	K	Joback Method
vc	1.734	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1570.53	J/mol×K	977.98	Joback Method

cpg	1599.37	J/molxK	1020.49	Joback Method
cpg	1626.22	J/molxK	1063.00	Joback Method
cpg	1651.25	J/molxK	1105.51	Joback Method
cpg	1674.62	J/molxK	1148.01	Joback Method
cpg	1696.50	J/molxK	1190.52	Joback Method
cpg	1717.04	J/molxK	1233.03	Joback Method
dvisc	0.0004393	Paxs	488.68	Joback Method
dvisc	0.0001061	Paxs	570.23	Joback Method
dvisc	0.0000366	Paxs	651.78	Joback Method
dvisc	0.0000160	Paxs	733.33	Joback Method
dvisc	0.0000082	Paxs	814.88	Joback Method
dvisc	0.0000048	Paxs	896.43	Joback Method
dvisc	0.0000030	Paxs	977.98	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.53922e+01
Coeff. B	-6.53503e+03
Coeff. C	-1.47148e+02
Temperature range (K), min.	579.80
Temperature range (K), max.	795.42

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C593500&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C593500&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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