

# 5-Chlorovaleric acid, octadecyl ester

<b>Inchi:</b>	InChI=1S/C23H45ClO2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-19-22-26-23(25)20-17
<b>InchiKey:</b>	NIFZBQWHRAITQZ-UHFFFAOYSA-N
<b>Formula:</b>	C23H45ClO2
<b>SMILES:</b>	CCCCCCCCCCCCCCCCCOC(=O)CCCCCl
<b>Mol. weight [g/mol]:</b>	389.06

## Physical Properties

Property code	Value	Unit	Source
gf	-103.07	kJ/mol	Joback Method
hf	-778.59	kJ/mol	Joback Method
hfus	62.31	kJ/mol	Joback Method
hvap	80.33	kJ/mol	Joback Method
log10ws	-8.47		Crippen Method
logp	8.200		Crippen Method
mcvol	354.610	ml/mol	McGowan Method
pc	856.47	kPa	Joback Method
tb	839.36	K	Joback Method
tc	1027.66	K	Joback Method
tf	451.05	K	Joback Method
vc	1.397	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1129.17	J/molxK	839.36	Joback Method
cpg	1149.50	J/molxK	870.74	Joback Method
cpg	1168.70	J/molxK	902.13	Joback Method
cpg	1186.80	J/molxK	933.51	Joback Method
cpg	1203.85	J/molxK	964.89	Joback Method
cpg	1219.88	J/molxK	996.28	Joback Method
cpg	1234.92	J/molxK	1027.66	Joback Method
dvisc	0.0009116	Paxs	451.05	Joback Method
dvisc	0.0003940	Paxs	515.77	Joback Method
dvisc	0.0002053	Paxs	580.49	Joback Method

dvisc	0.0001219	Paxs	645.20	Joback Method
dvisc	0.0000796	Paxs	709.92	Joback Method
dvisc	0.0000558	Paxs	774.64	Joback Method
dvisc	0.0000414	Paxs	839.36	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292318&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292318&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/78-188-8/5-Chlorovaleric-acid-octadecyl-ester.pdf>

Generated by Cheméo on 2024-04-27 20:50:59.143964996 +0000 UTC m=+16540308.064542313.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.