

# 1,4-Di-t-butylperoxy pyromellitate

<b>Inchi:</b>	InChI=1S/C18H22O10/c1-17(2,3)27-25-15(23)11-7-10(14(21)22)12(8-9(11)13(19)20)16(
<b>InchiKey:</b>	MXFZJFCNTQTGHU-UHFFFAOYSA-N
<b>Formula:</b>	C18H22O10
<b>SMILES:</b>	CC(C)(C)OOC(=O)c1cc(C(=O)O)c(C(=O)OOC(C)(C)C)cc1C(=O)O
<b>Mol. weight [g/mol]:</b>	398.36
<b>CAS:</b>	82234-37-5

## Physical Properties

Property code	Value	Unit	Source
chs	-8566.10 ± 6.00	kJ/mol	NIST Webbook
gf	-1019.44	kJ/mol	Joback Method
hf	-1513.89	kJ/mol	Joback Method
hfs	-1661.80 ± 6.00	kJ/mol	NIST Webbook
hfus	39.75	kJ/mol	Joback Method
hvap	127.31	kJ/mol	Joback Method
log10ws	-4.97		Crippen Method
logp	2.857		Crippen Method
mcvol	282.220	ml/mol	McGowan Method
pc	1968.30	kPa	Joback Method
tb	1135.92	K	Joback Method
tc	1396.14	K	Joback Method
tf	771.72	K	Joback Method
vc	1.048	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	916.47	J/molxK	1135.92	Joback Method
cpg	921.08	J/molxK	1179.29	Joback Method
cpg	923.99	J/molxK	1222.66	Joback Method
cpg	925.22	J/molxK	1266.03	Joback Method
cpg	924.82	J/molxK	1309.40	Joback Method
cpg	922.80	J/molxK	1352.77	Joback Method
cpg	919.21	J/molxK	1396.14	Joback Method

dvisc	0.0000046	Paxs	771.72	Joback Method
dvisc	0.0000023	Paxs	832.42	Joback Method
dvisc	0.0000013	Paxs	893.12	Joback Method
dvisc	0.0000008	Paxs	953.82	Joback Method
dvisc	0.0000005	Paxs	1014.52	Joback Method
dvisc	0.0000003	Paxs	1075.22	Joback Method
dvisc	0.0000002	Paxs	1135.92	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C82234375&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C82234375&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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