

# 1H-Indene, 5-butyl-6-hexyl-2,3-dihydro-

<b>Other names:</b>	5-n-Butyl-6-n-hexyl-(2,3-dihydroindene) 5-n-Butyl-6-n-hexylindan
<b>Inchi:</b>	InChI=1S/C19H30/c1-3-5-7-8-11-17-15-19-13-9-12-18(19)14-16(17)10-6-4-2/h14-15H,3-
<b>InchiKey:</b>	ZJQLBKPUALAMNC-UHFFFAOYSA-N
<b>Formula:</b>	C19H30
<b>SMILES:</b>	CCCCCc1cc2c(cc1CCCC)CCC2
<b>Mol. weight [g/mol]:</b>	258.44
<b>CAS:</b>	55030-45-0

## Physical Properties

Property code	Value	Unit	Source
gf	261.08	kJ/mol	Joback Method
hf	-140.23	kJ/mol	Joback Method
hfus	34.90	kJ/mol	Joback Method
hvap	62.37	kJ/mol	Joback Method
log10ws	-6.67		Crippen Method
logp	5.641		Crippen Method
mvol	243.950	ml/mol	McGowan Method
pc	1487.29	kPa	Joback Method
tb	687.15	K	Joback Method
tc	884.36	K	Joback Method
tf	390.05	K	Joback Method
vc	0.950	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	694.83	J/mol×K	687.15	Joback Method
cpg	714.37	J/mol×K	720.02	Joback Method
cpg	732.88	J/mol×K	752.89	Joback Method
cpg	750.40	J/mol×K	785.75	Joback Method
cpg	767.01	J/mol×K	818.62	Joback Method
cpg	782.76	J/mol×K	851.49	Joback Method
cpg	797.73	J/mol×K	884.36	Joback Method

dvisc	0.0015952	Paxs	390.05	Joback Method
dvisc	0.0009732	Paxs	439.57	Joback Method
dvisc	0.0006562	Paxs	489.08	Joback Method
dvisc	0.0004758	Paxs	538.60	Joback Method
dvisc	0.0003641	Paxs	588.12	Joback Method
dvisc	0.0002905	Paxs	637.63	Joback Method
dvisc	0.0002394	Paxs	687.15	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55030450&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55030450&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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