

Cyclopropa[c]2h-1-benzopyran,3,4-dihydro-4-phe

Inchi:	InChI=1S/C16H14O/c1-2-6-12(7-3-1)16-10-13(16)11-17-15-9-5-4-8-14(15)16/h1-9,13H,1
InchiKey:	WARAYOPBTKJQFK-UHFFFAOYSA-N
Formula:	C16H14O
SMILES:	<chem>c1ccc(C23CC2COc2ccccc23)cc1</chem>
Mol. weight [g/mol]:	222.28
CAS:	6719-35-3

Physical Properties

Property code	Value	Unit	Source
gf	341.02	kJ/mol	Joback Method
hf	123.02	kJ/mol	Joback Method
hfus	24.94	kJ/mol	Joback Method
hvap	59.44	kJ/mol	Joback Method
log10ws	-3.75		Crippen Method
logp	3.385		Crippen Method
mcvol	172.930	ml/mol	McGowan Method
pc	2966.57	kPa	Joback Method
tb	660.22	K	Joback Method
tc	923.82	K	Joback Method
tf	425.31	K	Joback Method
vc	0.656	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	468.51	J/molxK	660.22	Joback Method
cpg	485.91	J/molxK	704.15	Joback Method
cpg	502.08	J/molxK	748.09	Joback Method
cpg	517.38	J/molxK	792.02	Joback Method
cpg	532.18	J/molxK	835.95	Joback Method
cpg	546.85	J/molxK	879.89	Joback Method
cpg	561.75	J/molxK	923.82	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C6719353&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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