

# Pyrene, 1,2,3,3a,4,5-hexahydro-

<b>Other names:</b>	1,2,3,3a,4,5-Hexahydropyrene 1,2,3,9,10,10a-Hexahydropyrene
<b>Inchi:</b>	InChI=1S/C16H16/c1-3-11-7-9-13-5-2-6-14-10-8-12(4-1)15(11)16(13)14/h1,3-4,7,9,14H.2
<b>InchiKey:</b>	AEFBNJXWRHSZGO-UHFFFAOYSA-N
<b>Formula:</b>	C16H16
<b>SMILES:</b>	<chem>c1cc2c3c4c(ccc3c1)CCCC4CC2</chem>
<b>Mol. weight [g/mol]:</b>	208.30
<b>CAS:</b>	5385-37-5

## Physical Properties

Property code	Value	Unit	Source
gf	403.22	kJ/mol	Joback Method
hf	185.56	kJ/mol	Joback Method
hfus	22.29	kJ/mol	Joback Method
hvap	57.25	kJ/mol	Joback Method
log10ws	-5.44		Crippen Method
logp	4.206		Crippen Method
mcvol	171.360	ml/mol	McGowan Method
pc	2684.64	kPa	Joback Method
tb	644.23	K	Joback Method
tc	892.20	K	Joback Method
tf	406.88	K	Joback Method
vc	0.660	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	458.16	J/molxK	644.23	Joback Method
cpg	535.12	J/molxK	850.88	Joback Method
cpg	521.78	J/molxK	809.55	Joback Method
cpg	507.60	J/molxK	768.22	Joback Method
cpg	492.39	J/molxK	726.89	Joback Method
cpg	475.98	J/molxK	685.56	Joback Method
cpg	547.81	J/molxK	892.20	Joback Method

dvisc	0.0013641	Paxs	644.23	Joback Method
dvisc	0.0014521	Paxs	604.67	Joback Method
dvisc	0.0015592	Paxs	565.11	Joback Method
dvisc	0.0016923	Paxs	525.56	Joback Method
dvisc	0.0018615	Paxs	486.00	Joback Method
dvisc	0.0020824	Paxs	446.44	Joback Method
dvisc	0.0023809	Paxs	406.88	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5385375&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5385375&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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