

# Benzenepentacarboxylic acid, pentamethyl ester

Inchi:	InChI=1S/C16H16O10/c1-22-12(17)7-6-8(13(18)23-2)10(15(20)25-4)11(16(21)26-5)9(7)
InchiKey:	UVKUTUMTHIGYKO-UHFFFAOYSA-N
Formula:	C16H16O10
SMILES:	<chem>COC(=O)c1cc(C(=O)OC)c(C(=O)OC)c(C(=O)OC)c1C(=O)OC</chem>
Mol. weight [g/mol]:	368.29
CAS:	3327-06-8

## Physical Properties

Property code	Value	Unit	Source
chs	-6794.70 ± 1.40	kJ/mol	NIST Webbook
gf	-1011.87	kJ/mol	Joback Method
hf	-1406.92	kJ/mol	Joback Method
hfs	-1788.20 ± 1.60	kJ/mol	NIST Webbook
hfus	43.62	kJ/mol	Joback Method
hsub	165.10 ± 0.80	kJ/mol	NIST Webbook
hsub	165.10 ± 0.80	kJ/mol	NIST Webbook
hvap	101.91	kJ/mol	Joback Method
log10ws	-2.68		Crippen Method
logp	0.620		Crippen Method
mcvol	249.740	ml/mol	McGowan Method
pc	1970.05	kPa	Joback Method
rinpol	2345.00		NIST Webbook
rinpol	2345.00		NIST Webbook
tb	993.53	K	Joback Method
tc	1221.44	K	Joback Method
tf	424.70 ± 0.10	K	NIST Webbook
vc	0.944	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	764.26	J/mol×K	1183.46	Joback Method
cpg	761.06	J/mol×K	1221.44	Joback Method
cpg	750.46	J/mol×K	993.53	Joback Method

cpg	756.96	J/mol×K	1031.52	Joback Method
cpg	761.66	J/mol×K	1069.50	Joback Method
cpg	764.49	J/mol×K	1107.49	Joback Method
cpg	765.38	J/mol×K	1145.47	Joback Method
dvisc	0.0000371	Paxs	993.53	Joback Method
dvisc	0.0000443	Paxs	945.84	Joback Method
dvisc	0.0001536	Paxs	707.38	Joback Method
dvisc	0.0001125	Paxs	755.07	Joback Method
dvisc	0.0000855	Paxs	802.76	Joback Method
dvisc	0.0000670	Paxs	850.45	Joback Method
dvisc	0.0000539	Paxs	898.15	Joback Method
hfust	38.00	kJ/mol	424.70	NIST Webbook
hsubt	160.00 ± 0.80	kJ/mol	401.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3327068&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3327068&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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