

# Benzenemethanamine, «alpha»-methyl-N-(1-phenylethyl)-

<b>Other names:</b>	Bis(«alpha»-methylbenzyl)amine Bis(«alpha»-phenylethyl)amine Dibenzylamine, «alpha», «alpha»'-dimethyl- N,N-Bis(1-phenylethyl)amine di-(1-Phenylethyl)-amine
<b>Inchi:</b>	InChI=1S/C16H19N/c1-13(15-9-5-3-6-10-15)17-14(2)16-11-7-4-8-12-16/h3-14,17H,1-2H3
<b>InchiKey:</b>	NXLACVVNHYYJN-UHFFFAOYSA-N
<b>Formula:</b>	C16H19N
<b>SMILES:</b>	CC(NC(C)c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	225.33
<b>CAS:</b>	10024-74-5

## Physical Properties

Property code	Value	Unit	Source
gf	393.17	kJ/mol	Joback Method
hf	142.40	kJ/mol	Joback Method
hfus	23.33	kJ/mol	Joback Method
hvap	61.42	kJ/mol	Joback Method
log10ws	-4.83		Crippen Method
logp	4.098		Crippen Method
mcvol	198.760	ml/mol	McGowan Method
pc	2349.64	kPa	Joback Method
tb	668.13	K	Joback Method
tc	905.79	K	Joback Method
tf	345.58	K	Joback Method
vc	0.739	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	528.61	J/mol×K	668.13	Joback Method
cpg	547.15	J/mol×K	707.74	Joback Method
cpg	564.27	J/mol×K	747.35	Joback Method
cpg	580.04	J/mol×K	786.96	Joback Method

cpg	594.57	J/mol×K	826.57	Joback Method
cpg	607.94	J/mol×K	866.18	Joback Method
cpg	620.23	J/mol×K	905.79	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10024745&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10024745&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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