

# 3-Heptafluorobutyryl-«delta»-camphor

<b>Other names:</b>	3-(Heptafluorobutyryl)-D-camphor Bicyclo[2.2.1]heptan-2-one, 3-(2,2,3,3,4,4,4-heptafluoro-1-oxobutyl)-1,7,7-trimethyl-, (1R)- 3-Heptafluorobutyryl-(+)-camphor (1R)-3-(2,2,3,3,4,4,4-heptafluoro-1-oxobutyl)-1,7,7-trimethylbicyclo[2.2.1]heptan-2-one
<b>Inchi:</b>	InChI=1S/C14H15F7O2/c1-10(2)6-4-5-11(10,3)8(22)7(6)9(23)12(15,16)13(17,18)14(19,20)
<b>InchiKey:</b>	PEWOESYEGLBLNR-UHFFFAOYSA-N
<b>Formula:</b>	C14H15F7O2
<b>SMILES:</b>	CC12CCC(C(C(=O)C(F)(F)C(F)(F)C(F)(F)F)C1=O)C2(C)C
<b>Mol. weight [g/mol]:</b>	348.26
<b>CAS:</b>	51800-99-8

## Physical Properties

Property code	Value	Unit	Source
gf	-1456.66	kJ/mol	Joback Method
hf	-1852.35	kJ/mol	Joback Method
hfus	16.16	kJ/mol	Joback Method
hvap	45.22	kJ/mol	Joback Method
log10ws	-4.36		Crippen Method
logp	4.030		Crippen Method
mcvol	201.930	ml/mol	McGowan Method
pc	1717.45	kPa	Joback Method
tb	635.50	K	Joback Method
tc	828.57	K	Joback Method
tf	448.76	K	Joback Method
vc	0.826	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	612.54	J/molxK	635.50	Joback Method
cpg	628.39	J/molxK	667.68	Joback Method
cpg	643.29	J/molxK	699.86	Joback Method
cpg	657.45	J/molxK	732.03	Joback Method
cpg	671.05	J/molxK	764.21	Joback Method

cpg	684.30	J/mol×K	796.39	Joback Method
cpg	697.41	J/mol×K	828.57	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	338.20	K	0.03	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C51800998&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C51800998&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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