

# Benzoic acid, 3,4-dimethyl-, ethyl ester

<b>Other names:</b>	Ethyl 3,4-dimethylbenzoate
<b>Inchi:</b>	InChI=1S/C11H14O2/c1-4-13-11(12)10-6-5-8(2)9(3)7-10/h5-7H,4H2,1-3H3
<b>InchiKey:</b>	GTXUIKILNLXONM-UHFFFAOYSA-N
<b>Formula:</b>	C11H14O2
<b>SMILES:</b>	CCOC(=O)c1ccc(C)c(C)c1
<b>Mol. weight [g/mol]:</b>	178.23
<b>CAS:</b>	33499-44-4

## Physical Properties

Property code	Value	Unit	Source
gf	-99.03	kJ/mol	Joback Method
hf	-301.58	kJ/mol	Joback Method
hfus	20.30	kJ/mol	Joback Method
hvap	52.84	kJ/mol	Joback Method
log10ws	-3.08		Crippen Method
logp	2.480		Crippen Method
mcvol	149.530	ml/mol	McGowan Method
pc	2693.00	kPa	Joback Method
rinpol	1385.00		NIST Webbook
rinpol	1432.00		NIST Webbook
rinpol	1385.00		NIST Webbook
rinpol	1432.00		NIST Webbook
tb	564.01	K	Joback Method
tc	775.37	K	Joback Method
tf	337.35	K	Joback Method
vc	0.568	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	345.82	J/mol×K	564.01	Joback Method
cpg	359.69	J/mol×K	599.24	Joback Method
cpg	372.86	J/mol×K	634.46	Joback Method
cpg	385.33	J/mol×K	669.69	Joback Method

cpg	397.11	J/mol×K	704.92	Joback Method
cpg	408.21	J/mol×K	740.15	Joback Method
cpg	418.63	J/mol×K	775.37	Joback Method
dvisc	0.0013555	Paxs	337.35	Joback Method
dvisc	0.0008295	Paxs	375.13	Joback Method
dvisc	0.0005553	Paxs	412.90	Joback Method
dvisc	0.0003976	Paxs	450.68	Joback Method
dvisc	0.0002998	Paxs	488.46	Joback Method
dvisc	0.0002354	Paxs	526.23	Joback Method
dvisc	0.0001909	Paxs	564.01	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C33499444&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C33499444&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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