

# 2-methyl-4-(2,6,6-trimethyl-2-cyclohexen-1-yl)-2-b

<b>Other names:</b>	2-Butenal, 2-methyl-4-(2,6,6-trimethyl-2-cyclohexen-1-yl)-
<b>Inchi:</b>	InChI=1S/C14H22O/c1-11(10-15)7-8-13-12(2)6-5-9-14(13,3)4/h6-7,10,13H,5,8-9H2,1-4H
<b>InchiKey:</b>	JJHZLPJQTHPGEI-UHFFFAOYSA-N
<b>Formula:</b>	C14H22O
<b>SMILES:</b>	CC(C=O)=CCC1C(C)=CCCC1(C)C
<b>Mol. weight [g/mol]:</b>	206.32
<b>CAS:</b>	68555-62-4

## Physical Properties

Property code	Value	Unit	Source
chl	-8420.70 ± 2.30	kJ/mol	NIST Webbook
gf	70.73	kJ/mol	Joback Method
hf	-214.91	kJ/mol	Joback Method
hfl	227.20 ± 2.30	kJ/mol	NIST Webbook
hfus	20.64	kJ/mol	Joback Method
hvap	53.44	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	3.904		Crippen Method
mcvol	190.230	ml/mol	McGowan Method
pc	2079.33	kPa	Joback Method
tb	591.68	K	Joback Method
tc	804.87	K	Joback Method
tf	310.82	K	Joback Method
vc	0.734	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	489.02	J/molxK	591.68	Joback Method
cpg	508.04	J/molxK	627.21	Joback Method
cpg	525.99	J/molxK	662.74	Joback Method
cpg	542.98	J/molxK	698.28	Joback Method
cpg	559.14	J/molxK	733.81	Joback Method
cpg	574.59	J/molxK	769.34	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C68555624&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C68555624&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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