

# 3-Buten-2-one, 4-(1,2,6,6-tetramethyl-2-cyclohexen-1-yl)-

Other names: 4-(1,2,6,6-Tetramethyl-2-cyclohexen-1-yl)-3-buten-2-one

6-Methyl-«alpha»-(E)-ionone

Inchi: InChI=1S/C14H22O/c1-11-7-6-9-13(3,4)14(11,5)10-8-12(2)15/h7-8,10H,6,9H2,1-5H3/b10

InchiKey: MOLZHSROFREFLG-CSKARUKUSA-N

Formula: C14H22O

SMILES: CC(=O)C=CC1(C)C(C)=CCCC1(C)C

Mol. weight [g/mol]: 206.32

CAS: 107815-05-4

## Physical Properties

Property code	Value	Unit	Source
gf	44.39	kJ/mol	Joback Method
hf	-216.88	kJ/mol	Joback Method
hfus	14.96	kJ/mol	Joback Method
hvap	52.23	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	3.904		Crippen Method
mcvol	190.230	ml/mol	McGowan Method
pc	2135.43	kPa	Joback Method
rinpol	1515.00		NIST Webbook
rinpol	1497.00		NIST Webbook
rinpol	1515.00		NIST Webbook
rinpol	1497.00		NIST Webbook
tb	597.25	K	Joback Method
tc	819.93	K	Joback Method
tf	356.61	K	Joback Method
vc	0.720	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	487.29	J/molxK	597.25	Joback Method
cpg	506.21	J/molxK	634.36	Joback Method
cpg	524.06	J/molxK	671.48	Joback Method

cpg	541.08	J/mol×K	708.59	Joback Method
cpg	557.51	J/mol×K	745.70	Joback Method
cpg	573.55	J/mol×K	782.81	Joback Method
cpg	589.46	J/mol×K	819.93	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C107815054&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C107815054&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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