

# 12-methyltridecanal

<b>Other names:</b>	Tridecanal, 12-methyl-
<b>Inchi:</b>	InChI=1S/C14H28O/c1-14(2)12-10-8-6-4-3-5-7-9-11-13-15/h13-14H,3-12H2,1-2H3
<b>InchiKey:</b>	OQWNKUAAZQSLNSR-UHFFFAOYSA-N
<b>Formula:</b>	C14H28O
<b>SMILES:</b>	CC(C)CCCCCCCCCCC=O
<b>Mol. weight [g/mol]:</b>	212.37
<b>CAS:</b>	75853-49-5

## Physical Properties

Property code	Value	Unit	Source
gf	-34.96	kJ/mol	Joback Method
hf	-423.15	kJ/mol	Joback Method
hfus	30.78	kJ/mol	Joback Method
hvap	53.09	kJ/mol	Joback Method
log10ws	-4.72		Crippen Method
logp	4.742		Crippen Method
mcvol	209.690	ml/mol	McGowan Method
pc	1625.91	kPa	Joback Method
rinpol	1576.00		NIST Webbook
rinpol	1576.00		NIST Webbook
ripol	1863.00		NIST Webbook
ripol	1871.00		NIST Webbook
ripol	1889.00		NIST Webbook
ripol	1883.00		NIST Webbook
ripol	1884.00		NIST Webbook
ripol	1883.00		NIST Webbook
ripol	1863.00		NIST Webbook
tb	567.94	K	Joback Method
tc	735.40	K	Joback Method
tf	274.54	K	Joback Method
vc	0.831	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	537.53	J/mol×K	567.94	Joback Method
cpg	616.41	J/mol×K	707.49	Joback Method
cpg	601.99	J/mol×K	679.58	Joback Method
cpg	586.91	J/mol×K	651.67	Joback Method
cpg	571.15	J/mol×K	623.76	Joback Method
cpg	554.70	J/mol×K	595.85	Joback Method
cpg	630.19	J/mol×K	735.40	Joback Method
dvisc	0.0001986	Paxs	567.94	Joback Method
dvisc	0.0002704	Paxs	519.04	Joback Method
dvisc	0.0003924	Paxs	470.14	Joback Method
dvisc	0.0006210	Paxs	421.24	Joback Method
dvisc	0.0011087	Paxs	372.34	Joback Method
dvisc	0.0023587	Paxs	323.44	Joback Method
dvisc	0.0065661	Paxs	274.54	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51267e+01
Coeff. B	-4.85293e+03
Coeff. C	-9.35630e+01
Temperature range (K), min.	420.60
Temperature range (K), max.	587.99

## Sources

**The Yaws Handbook of Vapor Pressure:**  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C75853495&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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