

# tolyaldehyde propylene glycol acetal

<b>Other names:</b>	4-methyl-2-(o-tolyl)-1,3-dioxolane
<b>Inchi:</b>	InChI=1S/C11H14O2/c1-8-3-5-10(6-4-8)11-12-7-9(2)13-11/h3-6,9,11H,7H2,1-2H3
<b>InchiKey:</b>	VPBUQMKUEPQXKR-UHFFFAOYSA-N
<b>Formula:</b>	C11H14O2
<b>SMILES:</b>	<chem>Cc1ccc(C2OCC(C)O2)cc1</chem>
<b>Mol. weight [g/mol]:</b>	178.23
<b>CAS:</b>	58244-29-4

## Physical Properties

Property code	Value	Unit	Source
gf	1.12	kJ/mol	Joback Method
hf	-269.17	kJ/mol	Joback Method
hfus	28.86	kJ/mol	Joback Method
hvap	51.99	kJ/mol	Joback Method
log10ws	-2.72		Crippen Method
logp	2.429		Crippen Method
mcvol	142.970	ml/mol	McGowan Method
pc	2989.32	kPa	Joback Method
rinpol	1376.00		NIST Webbook
tb	547.25	K	Joback Method
tc	779.61	K	Joback Method
tf	312.47	K	Joback Method
vc	0.525	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.04	J/molxK	547.25	Joback Method
cpg	427.86	J/molxK	740.88	Joback Method
cpg	414.43	J/molxK	702.16	Joback Method
cpg	399.97	J/molxK	663.43	Joback Method
cpg	384.44	J/molxK	624.70	Joback Method
cpg	367.81	J/molxK	585.98	Joback Method
cpg	440.30	J/molxK	779.61	Joback Method

dvisc	0.0003348	Paxs	547.25	Joback Method
dvisc	0.0004104	Paxs	508.12	Joback Method
dvisc	0.0005203	Paxs	468.99	Joback Method
dvisc	0.0006887	Paxs	429.86	Joback Method
dvisc	0.0009645	Paxs	390.73	Joback Method
dvisc	0.0014557	Paxs	351.60	Joback Method
dvisc	0.0024358	Paxs	312.47	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C58244294&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C58244294&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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