

# Pimelic acid, 2-(2-methoxyethyl)heptyl 4-methyl-2-pentyl ester

<b>Inchi:</b>	InChI=1S/C23H44O5/c1-6-7-9-12-21(15-16-26-5)18-27-22(24)13-10-8-11-14-23(25)28-2
<b>InchiKey:</b>	BEKBNVBFIPJUAC-UHFFFAOYSA-N
<b>Formula:</b>	C23H44O5
<b>SMILES:</b>	CCCCC(CCOC)COC(=O)CCCCC(=O)OC(C)CC(C)C
<b>Mol. weight [g/mol]:</b>	400.59

## Physical Properties

Property code	Value	Unit	Source
gf	-437.38	kJ/mol	Joback Method
hf	-1155.71	kJ/mol	Joback Method
hfus	51.52	kJ/mol	Joback Method
hvap	86.35	kJ/mol	Joback Method
log10ws	-5.89		Crippen Method
logp	5.691		Crippen Method
mvol	355.680	ml/mol	McGowan Method
pc	903.43	kPa	Joback Method
rinpol	2510.00		NIST Webbook
rinpol	2510.00		NIST Webbook
tb	899.32	K	Joback Method
tc	1101.14	K	Joback Method
tf	470.52	K	Joback Method
vc	1.371	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1187.95	J/molxK	899.32	Joback Method
cpg	1207.15	J/molxK	932.96	Joback Method
cpg	1224.89	J/molxK	966.59	Joback Method
cpg	1241.18	J/molxK	1000.23	Joback Method
cpg	1256.04	J/molxK	1033.86	Joback Method
cpg	1269.50	J/molxK	1067.50	Joback Method
cpg	1281.56	J/molxK	1101.14	Joback Method
dvisc	0.0006365	Paxs	470.52	Joback Method

dvisc	0.0002422	Paxs	541.99	Joback Method
dvisc	0.0001154	Paxs	613.45	Joback Method
dvisc	0.0000642	Paxs	684.92	Joback Method
dvisc	0.0000399	Paxs	756.39	Joback Method
dvisc	0.0000269	Paxs	827.85	Joback Method
dvisc	0.0000193	Paxs	899.32	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U406733&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U406733&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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