

# Dihydroconiferyl alcohol

<b>Other names:</b>	guaiacyl propanol 3-(4-hydroxy-3-methoxyphenyl)-propan-1-ol 3-(4-guaiacyl)propanol 3-(4-hydroxy-3-methoxy-phenyl)-1-propanol 1-Propanol, 3-(4-hydroxy-3-methoxyphenyl)
<b>Inchi:</b>	InChI=1S/C10H14O3/c1-13-10-7-8(3-2-6-11)4-5-9(10)12/h4-5,7,11-12H,2-3,6H2,1H3
<b>InchiKey:</b>	MWOMNLDJNQWJMK-UHFFFAOYSA-N
<b>Formula:</b>	C10H14O3
<b>SMILES:</b>	COc1cc(CCCO)ccc1O
<b>Mol. weight [g/mol]:</b>	182.22
<b>CAS:</b>	2305-13-7

## Physical Properties

Property code	Value	Unit	Source
gf	-260.34	kJ/mol	Joback Method
hf	-486.43	kJ/mol	Joback Method
hfus	26.37	kJ/mol	Joback Method
hvap	72.89	kJ/mol	Joback Method
log10ws	-1.62		Crippen Method
logp	1.326		Crippen Method
mcvol	145.610	ml/mol	McGowan Method
pc	3727.11	kPa	Joback Method
ripol	1660.60		NIST Webbook
ripol	1597.00		NIST Webbook
ripol	2941.00		NIST Webbook
ripol	2970.00		NIST Webbook
ripol	2970.00		NIST Webbook
ripol	2969.00		NIST Webbook
ripol	2969.00		NIST Webbook
ripol	2980.00		NIST Webbook
ripol	2980.00		NIST Webbook
tb	655.08	K	Joback Method
tc	857.18	K	Joback Method
tf	436.17	K	Joback Method
vc	0.490	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	384.34	J/molxK	655.08	Joback Method
cpg	395.11	J/molxK	688.76	Joback Method
cpg	405.31	J/molxK	722.45	Joback Method
cpg	414.98	J/molxK	756.13	Joback Method
cpg	424.16	J/molxK	789.81	Joback Method
cpg	432.90	J/molxK	823.50	Joback Method
cpg	441.25	J/molxK	857.18	Joback Method
dvisc	0.0004872	Paxs	436.17	Joback Method
dvisc	0.0001812	Paxs	472.66	Joback Method
dvisc	0.0000777	Paxs	509.14	Joback Method
dvisc	0.0000373	Paxs	545.62	Joback Method
dvisc	0.0000196	Paxs	582.11	Joback Method
dvisc	0.0000111	Paxs	618.60	Joback Method
dvisc	0.0000067	Paxs	655.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2305137&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2305137&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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