

# Ethanol, 2-[methyl(phenylmethyl)amino]-

<b>Other names:</b>	Ethanol, 2-(benzylmethylamino)- N-Methyl-N-benzylethanolamine 2-(Benzylmethylamino)ethanol N-Benzyl-N-methylethanolamine N-Benzyl-N-methyl-N-ethanolamine
<b>Inchi:</b>	InChI=1S/C10H15NO/c1-11(7-8-12)9-10-5-3-2-4-6-10/h2-6,12H,7-9H2,1H3
<b>InchiKey:</b>	WOUANPHGFPAJCA-UHFFFAOYSA-N
<b>Formula:</b>	C10H15NO
<b>SMILES:</b>	CN(CCO)Cc1ccccc1
<b>Mol. weight [g/mol]:</b>	165.23
<b>CAS:</b>	101-98-4

## Physical Properties

Property code	Value	Unit	Source
gf	119.69	kJ/mol	Joback Method
hf	-97.90	kJ/mol	Joback Method
hfus	22.81	kJ/mol	Joback Method
hvap	58.85	kJ/mol	Joback Method
log10ws	-1.44		Crippen Method
logp	1.111		Crippen Method
mcvol	143.850	ml/mol	McGowan Method
pc	3239.34	kPa	Joback Method
rinpol	1328.70		NIST Webbook
tb	559.50	K	Joback Method
tc	749.63	K	Joback Method
tf	322.17	K	Joback Method
vc	0.524	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	346.59	J/mol×K	559.50	Joback Method
cpg	359.68	J/mol×K	591.19	Joback Method
cpg	372.02	J/mol×K	622.88	Joback Method

cpg	383.64	J/mol×K	654.56	Joback Method
cpg	394.58	J/mol×K	686.25	Joback Method
cpg	404.87	J/mol×K	717.94	Joback Method
cpg	414.55	J/mol×K	749.63	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C101984&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C101984&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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