

# n-Heptyl acrylate

<b>Other names:</b>	2-Propenoic acid, heptyl ester Acrylic acid, heptyl ester ENT 15748 Heptyl acrylate
<b>Inchi:</b>	InChI=1S/C10H18O2/c1-3-5-6-7-8-9-12-10(11)4-2/h4H,2-3,5-9H2,1H3
<b>InchiKey:</b>	SCFQUKBBGYTJNC-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O2
<b>SMILES:</b>	<chem>C=CC(=O)OCCCCCCC</chem>
<b>Mol. weight [g/mol]:</b>	170.25
<b>CAS:</b>	2499-58-3

## Physical Properties

Property code	Value	Unit	Source
gf	-112.76	kJ/mol	Joback Method
hf	-369.10	kJ/mol	Joback Method
hfus	23.16	kJ/mol	Joback Method
hvap	46.34	kJ/mol	Joback Method
log10ws	-2.72		Crippen Method
logp	2.686		Crippen Method
mcvol	154.900	ml/mol	McGowan Method
pc	2289.32	kPa	Joback Method
rinpol	1173.00		NIST Webbook
ripol	1489.00		NIST Webbook
tb	501.17	K	Joback Method
tc	676.68	K	Joback Method
tf	272.86	K	Joback Method
vc	0.601	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	349.85	J/mol×K	501.17	Joback Method
cpg	363.46	J/mol×K	530.42	Joback Method
cpg	376.52	J/mol×K	559.67	Joback Method

cpg	389.06	J/mol×K	588.93	Joback Method
cpg	401.07	J/mol×K	618.18	Joback Method
cpg	412.57	J/mol×K	647.43	Joback Method
cpg	423.57	J/mol×K	676.68	Joback Method
dvisc	0.0030420	Paxs	272.86	Joback Method
dvisc	0.0015143	Paxs	310.91	Joback Method
dvisc	0.0008777	Paxs	348.96	Joback Method
dvisc	0.0005663	Paxs	387.01	Joback Method
dvisc	0.0003952	Paxs	425.07	Joback Method
dvisc	0.0002926	Paxs	463.12	Joback Method
dvisc	0.0002268	Paxs	501.17	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.57305e+01
Coeff. B	-4.51965e+03
Coeff. C	-7.63250e+01
Temperature range (K), min.	368.99
Temperature range (K), max.	510.11

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2499583&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2499583&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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