

1,1,1-trifluorodecane

Other names:	Decane, 1,1,1-trifluoro
Inchi:	InChI=1S/C10H19F3/c1-2-3-4-5-6-7-8-9-10(11,12)13/h2-9H2,1H3
InchiKey:	IGGXVAUUTYVHBN-UHFFFAOYSA-N
Formula:	C10H19F3
SMILES:	CCCCCCCCCCC(F)(F)F
Mol. weight [g/mol]:	196.25
CAS:	26288-16-4

Physical Properties

Property code	Value	Unit	Source
af	0.5405		Cheméo Relay (1.0)
gf	-548.27	kJ/mol	Joback Method
hf	-899.29	kJ/mol	Cheméo Relay (1.0)
hfus	23.48	kJ/mol	Joback Method
hvap	45.30	kJ/mol	Cheméo Relay (1.0)
ie	10.81	eV	Cheméo Relay (1.0)
log10ws	-5.24		Cheméo Relay (1.0)
logp	4.689		Crippen Method
mcvol	157.070	ml/mol	McGowan Method
pc	1900.26	kPa	Joback Method
tb	424.69	K	Cheméo Relay (1.0)
tc	559.29	K	Cheméo Relay (1.0)
tf	209.09	K	Cheméo Relay (1.0)
vc	0.610	m3/kmol	Cheméo Relay (1.0)

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	341.37	J/mol×K	422.78	Joback Method
cpg	355.77	J/mol×K	447.54	Joback Method
cpg	369.59	J/mol×K	472.30	Joback Method
cpg	382.84	J/mol×K	497.07	Joback Method
cpg	395.54	J/mol×K	521.83	Joback Method
cpg	407.70	J/mol×K	546.59	Joback Method

cpg

419.36

J/mol×K

571.35

Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.29907e+01
Coeff. B	-3.38340e+03
Coeff. C	-6.06000e+01
Temperature range (K), min.	326.94
Temperature range (K), max.	501.19

Sources

Joback Method:https://en.wikipedia.org/wiki/Joback_method**McGowan Method:**<http://link.springer.com/article/10.1007/BF02311772>**Crippen Method:**<http://pubs.acs.org/doi/abs/10.1021/ci990307l>**Crippen Method:**https://www.chemeo.com/doc/models/crippen_log10ws**Cheméo Relay (1.0):****Cheméo Relay (1.0, beta):****The Yaws Handbook of Vapor Pressure:
NIST Webbook:**<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure><http://webbook.nist.gov/cgi/cbook.cgi?ID=C26288164&Units=SI>

Legend

af: Acentric Factor**cpg:** Ideal gas heat capacity**gf:** Standard Gibbs free energy of formation**hf:** Enthalpy of formation at standard conditions**hfus:** Enthalpy of fusion at standard conditions**hvap:** Enthalpy of vaporization at standard conditions**ie:** Ionization energy**log10ws:** Log10 of Water solubility in mol/l**logp:** Octanol/Water partition coefficient

mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/78-746-8/1-1-1-trifluorodecane.pdf>

Generated by Cheméo on 2026-07-21 17:16:46.183262416 +0000 UTC m=+165302.025147796.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.