

# ethyl 3-hydroxyoctanoate

<b>Inchi:</b>	InChI=1S/C10H20O3/c1-3-5-6-7-9(11)8-10(12)13-4-2/h9,11H,3-8H2,1-2H3
<b>InchiKey:</b>	VGWUJHSTGYCXQQ-UHFFFAOYSA-N
<b>Formula:</b>	C10H20O3
<b>SMILES:</b>	CCCCC(O)CC(=O)OCC
<b>Mol. weight [g/mol]:</b>	188.26
<b>CAS:</b>	7367-90-0

## Physical Properties

Property code	Value	Unit	Source
gf	-339.86	kJ/mol	Joback Method
hf	-652.04	kJ/mol	Joback Method
hfus	25.01	kJ/mol	Joback Method
hvap	63.30	kJ/mol	Joback Method
log10ws	-2.25		Crippen Method
logp	1.881		Crippen Method
mcvol	165.070	ml/mol	McGowan Method
pc	2421.88	kPa	Joback Method
ripol	1884.00		NIST Webbook
ripol	1892.00		NIST Webbook
ripol	1886.00		NIST Webbook
ripol	1875.00		NIST Webbook
ripol	1875.00		NIST Webbook
ripol	1886.00		NIST Webbook
ripol	1874.00		NIST Webbook
ripol	1880.00		NIST Webbook
ripol	1880.00		NIST Webbook
ripol	1886.00		NIST Webbook
tb	596.23	K	Joback Method
tc	765.82	K	Joback Method
tf	320.44	K	Joback Method
vc	0.632	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	428.91	J/mol×K	596.23	Joback Method
cpg	441.47	J/mol×K	624.50	Joback Method
cpg	453.52	J/mol×K	652.76	Joback Method
cpg	465.05	J/mol×K	681.03	Joback Method
cpg	476.07	J/mol×K	709.29	Joback Method
cpg	486.60	J/mol×K	737.56	Joback Method
cpg	496.65	J/mol×K	765.82	Joback Method
dvisc	0.0094549	Paxs	320.44	Joback Method
dvisc	0.0025198	Paxs	366.41	Joback Method
dvisc	0.0009018	Paxs	412.37	Joback Method
dvisc	0.0003966	Paxs	458.34	Joback Method
dvisc	0.0002026	Paxs	504.30	Joback Method
dvisc	0.0001158	Paxs	550.27	Joback Method
dvisc	0.0000721	Paxs	596.23	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7367900&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7367900&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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