

5-heptyl-dihydrofuran-2(3H)-thione

Inchi:	InChI=1S/C11H18OS/c1-2-3-4-5-6-7-10-8-9-11(13)12-10/h8H,2-7,9H2,1H3
InchiKey:	NLGVCAHTJNYDCN-UHFFFAOYSA-N
Formula:	C11H18OS
SMILES:	CCCCCCCC1=CCC(=S)O1
Mol. weight [g/mol]:	198.32

Physical Properties

Property code	Value	Unit	Source
gf	111.06	kJ/mol	Joback Method
hf	-160.14	kJ/mol	Joback Method
hfus	31.96	kJ/mol	Joback Method
hvap	53.59	kJ/mol	Joback Method
log10ws	-4.61		Crippen Method
logp	3.978		Crippen Method
mcvol	168.610	ml/mol	McGowan Method
pc	2537.93	kPa	Joback Method
ripol	2490.00		NIST Webbook
tb	574.76	K	Joback Method
tc	782.28	K	Joback Method
tf	332.39	K	Joback Method
vc	0.638	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	405.42	J/mol×K	574.76	Joback Method
cpg	420.26	J/mol×K	609.35	Joback Method
cpg	434.23	J/mol×K	643.93	Joback Method
cpg	447.38	J/mol×K	678.52	Joback Method
cpg	459.78	J/mol×K	713.11	Joback Method
cpg	471.47	J/mol×K	747.69	Joback Method
cpg	482.50	J/mol×K	782.28	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R422496&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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