

# 2-Naphthalenemethanol

**Inchi:** InChI=1S/C11H10O/c12-8-9-5-6-10-3-1-2-4-11(10)7-9/h1-7,12H,8H2  
**InchiKey:** MFGWMAAZYZSWMY-UHFFFAOYSA-N  
**Formula:** C11H10O  
**SMILES:** OCc1ccc2ccccc2c1  
**Mol. weight [g/mol]:** 158.20  
**CAS:** 1592-38-7

## Physical Properties

Property code	Value	Unit	Source
gf	114.35	kJ/mol	Joback Method
hf	-6.47	kJ/mol	Joback Method
hfus	19.00	kJ/mol	Joback Method
hsub	106.00 ± 2.10	kJ/mol	NIST Webbook
hvap	61.34	kJ/mol	Joback Method
log10ws	-3.42		Crippen Method
logp	2.332		Crippen Method
mvol	128.500	ml/mol	McGowan Method
pc	3805.69	kPa	Joback Method
tb	593.90	K	Joback Method
tc	810.04	K	Joback Method
tf	346.19	K	Joback Method
vc	0.484	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.08	J/molxK	593.90	Joback Method
cpg	312.40	J/molxK	629.92	Joback Method
cpg	322.92	J/molxK	665.95	Joback Method
cpg	332.70	J/molxK	701.97	Joback Method
cpg	341.80	J/molxK	738.00	Joback Method
cpg	350.29	J/molxK	774.02	Joback Method
cpg	358.22	J/molxK	810.04	Joback Method
dvisc	0.0035979	Paxs	346.19	Joback Method

dvisc	0.0015432	Paxs	387.48	Joback Method
dvisc	0.0007791	Paxs	428.76	Joback Method
dvisc	0.0004435	Paxs	470.04	Joback Method
dvisc	0.0002765	Paxs	511.33	Joback Method
dvisc	0.0001850	Paxs	552.62	Joback Method
dvisc	0.0001309	Paxs	593.90	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1592387&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1592387&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/79-082-4/2-Naphthalenemethanol.pdf>

Generated by Cheméo on 2024-04-29 01:42:45.186000306 +0000 UTC m=+16644214.106577634.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.