

# Dihydroartemisinic aldehyde, isomer # 2

<b>Inchi:</b>	InChI=1S/C15H24O/c1-10-4-6-13-11(2)5-7-14(12(3)9-16)15(13)8-10/h8-9,11-15H,4-7H2
<b>InchiKey:</b>	PIUSZJFEZXYOAT-OKMQTDMMSA-N
<b>Formula:</b>	C15H24O
<b>SMILES:</b>	CC1=CC2C(C(C)C=O)CCC(C)C2CC1
<b>Mol. weight [g/mol]:</b>	220.35

## Physical Properties

Property code	Value	Unit	Source
gf	51.47	kJ/mol	Joback Method
hf	-317.20	kJ/mol	Joback Method
hfus	24.22	kJ/mol	Joback Method
hvap	56.17	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	3.840		Crippen Method
mcvol	197.760	ml/mol	McGowan Method
pc	1956.14	kPa	Joback Method
ripol	2106.00		NIST Webbook
tb	616.18	K	Joback Method
tc	831.10	K	Joback Method
tf	312.41	K	Joback Method
vc	0.752	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	554.04	J/molxK	616.18	Joback Method
cpg	576.39	J/molxK	652.00	Joback Method
cpg	597.38	J/molxK	687.82	Joback Method
cpg	617.07	J/molxK	723.64	Joback Method
cpg	635.50	J/molxK	759.46	Joback Method
cpg	652.72	J/molxK	795.28	Joback Method
cpg	668.77	J/molxK	831.10	Joback Method
dvisc	0.0029027	Paxs	312.41	Joback Method
dvisc	0.0017507	Paxs	363.04	Joback Method

dvisc	0.0011950	Paxs	413.67	Joback Method
dvisc	0.0008866	Paxs	464.30	Joback Method
dvisc	0.0006975	Paxs	514.92	Joback Method
dvisc	0.0005728	Paxs	565.55	Joback Method
dvisc	0.0004859	Paxs	616.18	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R603848&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R603848&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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