

# 3,4-Dihydro-2H-pyran, 3-isopropyl-6-methyl-2-(3-methylcyclopent-2-enyl

Inchi:	InChI=1S/C15H24O/c1-10(2)14-8-6-12(4)16-15(14)13-7-5-11(3)9-13/h6,9-10,13-15H,5,7
InchiKey:	GUDFQP BIOYFJEG-UHFFFAOYSA-N
Formula:	C15H24O
SMILES:	CC1=CC(C2OC(C)=CCC2C(C)C)CC1
Mol. weight [g/mol]:	220.35

## Physical Properties

Property code	Value	Unit	Source
gf	80.81	kJ/mol	Joback Method
hf	-303.13	kJ/mol	Joback Method
hfus	27.57	kJ/mol	Joback Method
hvap	55.39	kJ/mol	Joback Method
log10ws	-4.57		Crippen Method
logp	4.308		Crippen Method
mcvol	197.760	ml/mol	McGowan Method
pc	1949.23	kPa	Joback Method
rinpol	1478.00		NIST Webbook
ripol	1723.00		NIST Webbook
tb	607.55	K	Joback Method
tc	828.08	K	Joback Method
tf	310.98	K	Joback Method
vc	0.736	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	541.33	J/molxK	607.55	Joback Method
cpg	641.87	J/molxK	791.32	Joback Method
cpg	624.35	J/molxK	754.57	Joback Method
cpg	605.57	J/molxK	717.81	Joback Method
cpg	585.51	J/molxK	681.06	Joback Method
cpg	564.11	J/molxK	644.30	Joback Method
cpg	658.18	J/molxK	828.08	Joback Method
dvisc	0.0002660	Paxs	607.55	Joback Method

dvisc	0.0003301	Paxs	558.12	Joback Method
dvisc	0.0004271	Paxs	508.69	Joback Method
dvisc	0.0005843	Paxs	459.26	Joback Method
dvisc	0.0008621	Paxs	409.84	Joback Method
dvisc	0.0014150	Paxs	360.41	Joback Method
dvisc	0.0027190	Paxs	310.98	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R198757&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R198757&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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