

# Cedroxyde

<b>Other names:</b>	1,5,9-trimethyl-13-oxabicyclo[10.1.0]trideca-4,8-diene
<b>Inchi:</b>	InChI=1S/C15H24O/c1-12-6-4-7-13(2)9-10-14-15(3,16-14)11-5-8-12/h7-8,14H,4-6,9-11H
<b>InchiKey:</b>	PAZWFUGWOAQBJJ-SWZPTJTJSA-N
<b>Formula:</b>	C15H24O
<b>SMILES:</b>	CC1=CCCC2(C)OC2CCC(C)=CCC1
<b>Mol. weight [g/mol]:</b>	220.35
<b>CAS:</b>	13786-79-3

## Physical Properties

Property code	Value	Unit	Source
gf	61.27	kJ/mol	Joback Method
hf	-274.59	kJ/mol	Joback Method
hfus	19.52	kJ/mol	Joback Method
hvap	55.28	kJ/mol	Joback Method
log10ws	-4.91		Crippen Method
logp	4.391		Crippen Method
mcvol	197.760	ml/mol	McGowan Method
pc	2173.43	kPa	Joback Method
rinpol	1714.00		NIST Webbook
tb	621.44	K	Joback Method
tc	861.53	K	Joback Method
tf	347.08	K	Joback Method
vc	0.725	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	541.92	J/molxK	621.44	Joback Method
cpg	565.74	J/molxK	661.45	Joback Method
cpg	588.08	J/molxK	701.47	Joback Method
cpg	609.09	J/molxK	741.48	Joback Method
cpg	628.92	J/molxK	781.50	Joback Method
cpg	647.73	J/molxK	821.51	Joback Method
cpg	665.68	J/molxK	861.53	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13786793&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13786793&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/79-168-9/Cedroxyde.pdf>

Generated by Cheméo on 2024-04-27 18:32:54.939636425 +0000 UTC m=+16532023.860213747.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.