

# 2-(Methylamino)ethanol, N,O-bis(trimethylsilyl)

**Other names:**

Ethanol, 2-(methylamino)-, N,O-bis(trimethylsilyl)  
N,1,1,1-Tetramethyl-N-(2-(((trimethylsilyl)oxy)ethyl)silanamine  
Sesquicineole

1,3-Dimethyl-3-(4-methylpent-3-en-1-yl)-2-oxabicyclo[2.2.2]octane

**Inchi:** InChI=1S/C15H26O/c1-12(2)6-5-9-15(4)13-7-10-14(3,16-15)11-8-13/h6,13H,5,7-11H2,1-**InchiKey:** BAQRIYKLDIPFQB-UHFFFAOYSA-N**Formula:** C15H26O**SMILES:** CC(C)=CCCC1(C)OC2(C)CCC1CC2**Mol. weight [g/mol]:** 222.37**CAS:** 90131-02-5

## Physical Properties

Property code	Value	Unit	Source
gf	139.58	kJ/mol	Joback Method
hf	-234.08	kJ/mol	Joback Method
hfus	22.02	kJ/mol	Joback Method
hvap	51.09	kJ/mol	Joback Method
log10ws	-4.81		Crippen Method
logp	4.471		Crippen Method
mcvol	202.060	ml/mol	McGowan Method
pc	1987.66	kPa	Joback Method
rinpol	1516.00		NIST Webbook
rinpol	1516.00		NIST Webbook
rinpol	1524.00		NIST Webbook
rinpol	1514.00		NIST Webbook
rinpol	1516.00		NIST Webbook
rinpol	1517.00		NIST Webbook
rinpol	1515.00		NIST Webbook
ripol	1715.00		NIST Webbook
ripol	1715.00		NIST Webbook
tb	591.42	K	Joback Method
tc	809.30	K	Joback Method
tf	338.74	K	Joback Method
vc	0.770	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	551.75	J/mol×K	591.42	Joback Method
cpg	573.41	J/mol×K	627.73	Joback Method
cpg	593.76	J/mol×K	664.05	Joback Method
cpg	613.05	J/mol×K	700.36	Joback Method
cpg	631.55	J/mol×K	736.67	Joback Method
cpg	649.52	J/mol×K	772.99	Joback Method
cpg	667.22	J/mol×K	809.30	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C90131025&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C90131025&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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