

# 13-methyltetradecanol

<b>Inchi:</b>	InChI=1S/C15H32O/c1-15(2)13-11-9-7-5-3-4-6-8-10-12-14-16/h15-16H,3-14H2,1-2H3
<b>InchiKey:</b>	FDAZSZUYCOPJED-UHFFFAOYSA-N
<b>Formula:</b>	C15H32O
<b>SMILES:</b>	CC(C)CCCCCCCCCCCCO
<b>Mol. weight [g/mol]:</b>	228.41

## Physical Properties

Property code	Value	Unit	Source
gf	-63.84	kJ/mol	Joback Method
hf	-510.44	kJ/mol	Joback Method
hfus	35.17	kJ/mol	Joback Method
hvap	65.27	kJ/mol	Joback Method
log10ws	-5.12		Crippen Method
logp	4.926		Crippen Method
mcvol	228.080	ml/mol	McGowan Method
pc	1528.27	kPa	Joback Method
rinpol	1740.00		NIST Webbook
rinpol	1736.00		NIST Webbook
tb	634.34	K	Joback Method
tc	794.88	K	Joback Method
tf	304.63	K	Joback Method
vc	0.888	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	637.33	J/molxK	634.34	Joback Method
cpg	653.95	J/molxK	661.10	Joback Method
cpg	669.89	J/molxK	687.85	Joback Method
cpg	685.17	J/molxK	714.61	Joback Method
cpg	699.82	J/molxK	741.36	Joback Method
cpg	713.84	J/molxK	768.12	Joback Method
cpg	727.26	J/molxK	794.88	Joback Method
dvisc	0.0175571	Paxs	304.63	Joback Method

dvisc	0.0030635	Paxs	359.58	Joback Method
dvisc	0.0008492	Paxs	414.53	Joback Method
dvisc	0.0003179	Paxs	469.49	Joback Method
dvisc	0.0001462	Paxs	524.44	Joback Method
dvisc	0.0000779	Paxs	579.39	Joback Method
dvisc	0.0000463	Paxs	634.34	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R213559&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R213559&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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