

# 8-Pentadecanone

<b>Other names:</b>	8-Oxopentadecane Caprylone Diheptyl ketone Heptyl ketone Pentadecan-8-one di-n-Heptyl ketone pentadecane-8-one
<b>Inchi:</b>	InChI=1S/C15H30O/c1-3-5-7-9-11-13-15(16)14-12-10-8-6-4-2/h3-14H2,1-2H3
<b>InchiKey:</b>	PQYGSSYFJIJDFK-UHFFFAOYSA-N
<b>Formula:</b>	C15H30O
<b>SMILES:</b>	CCCCCCCC(=O)CCCCCCC
<b>Mol. weight [g/mol]:</b>	226.40
<b>CAS:</b>	818-23-5

## Physical Properties

Property code	Value	Unit	Source
gf	-53.50	kJ/mol	Joback Method
hf	-465.51	kJ/mol	Joback Method
hfus	36.20	kJ/mol	Joback Method
hvap	55.73	kJ/mol	Joback Method
log10ws	-5.38		Crippen Method
logp	5.277		Crippen Method
mvol	223.780	ml/mol	McGowan Method
pc	1485.00	kPa	Joback Method
rinpol	1647.00		NIST Webbook
rinpol	1659.00		NIST Webbook
rinpol	1647.00		NIST Webbook
tb	596.47	K	Joback Method
tc	763.83	K	Joback Method
tf	315.15 ± 1.00	K	NIST Webbook
vc	0.881	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	671.69	J/mol×K	735.93	Joback Method
cpg	625.16	J/mol×K	652.26	Joback Method
cpg	608.23	J/mol×K	624.36	Joback Method
cpg	590.56	J/mol×K	596.47	Joback Method
cpg	641.36	J/mol×K	680.15	Joback Method
cpg	685.85	J/mol×K	763.83	Joback Method
cpg	656.87	J/mol×K	708.04	Joback Method
dvisc	0.0037799	Paxs	308.74	Joback Method
dvisc	0.0001743	Paxs	596.47	Joback Method
dvisc	0.0002327	Paxs	548.52	Joback Method
dvisc	0.0003282	Paxs	500.56	Joback Method
dvisc	0.0004978	Paxs	452.61	Joback Method
dvisc	0.0008337	Paxs	404.65	Joback Method
dvisc	0.0016037	Paxs	356.69	Joback Method
hvapt	53.00	kJ/mol	517.00	NIST Webbook
hvapt	65.40	kJ/mol	516.00	NIST Webbook
hvapt	65.30	kJ/mol	505.50	NIST Webbook
hvapt	61.90	kJ/mol	450.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51741e+01
Coeff. B	-4.93744e+03
Coeff. C	-9.84280e+01
Temperature range (K), min.	430.10
Temperature range (K), max.	599.05

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C818235&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C818235&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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