

9H-Fluorene, 1,9-dimethyl-

Other names:	1,9-Dimethylfluorene Fluorene, 1,9-dimethyl-
Inchi:	InChI=1S/C15H14/c1-10-6-5-9-14-13-8-4-3-7-12(13)11(2)15(10)14/h3-9,11H,1-2H3
InchiKey:	BNGCFLDEAHJKPE-UHFFFAOYSA-N
Formula:	C15H14
SMILES:	<chem>Cc1cccc2c1C(C)c1cccc1-2</chem>
Mol. weight [g/mol]:	194.27
CAS:	17057-98-6

Physical Properties

Property code	Value	Unit	Source
gf	356.30	kJ/mol	Joback Method
hf	170.84	kJ/mol	Joback Method
hfus	23.86	kJ/mol	Joback Method
hvap	55.09	kJ/mol	Joback Method
log10ws	-5.28		Crippen Method
logp	4.127		Crippen Method
mvol	163.830	ml/mol	McGowan Method
pc	2600.43	kPa	Joback Method
tb	609.10	K	Joback Method
tc	847.49	K	Joback Method
tf	374.19	K	Joback Method
vc	0.632	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	403.25	J/molxK	609.10	Joback Method
cpg	473.18	J/molxK	807.76	Joback Method
cpg	461.12	J/molxK	768.03	Joback Method
cpg	448.21	J/molxK	728.29	Joback Method
cpg	434.34	J/molxK	688.56	Joback Method
cpg	419.39	J/molxK	648.83	Joback Method
cpg	484.50	J/molxK	847.49	Joback Method

dvisc	0.0006520	Paxs	609.10	Joback Method
dvisc	0.0007050	Paxs	569.95	Joback Method
dvisc	0.0007712	Paxs	530.80	Joback Method
dvisc	0.0008558	Paxs	491.65	Joback Method
dvisc	0.0009668	Paxs	452.49	Joback Method
dvisc	0.0011178	Paxs	413.34	Joback Method
dvisc	0.0013323	Paxs	374.19	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.02879e+01
Coeff. B	-2.93205e+03
Coeff. C	-8.67250e+01
Temperature range (K), min.	379.92
Temperature range (K), max.	675.91

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C17057986&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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