

# 5H-Dibenzo[a,d]cyclohepten-5-ol, 10,11-dihydro-

Other names:

Dibenzosuberol  
10,11-Dihydro-5H-dibenzo(a,d)cyclohepten-5-ol  
5-Hydroxydibenzo[a,d]cyclohepta[1,4]diene  
dibenzo(b,f)cycloheptan-1-ol

Inchi:

InChI=1S/C15H14O/c16-15-13-7-3-1-5-11(13)9-10-12-6-2-4-8-14(12)15/h1-8,15-16H,9-1

InchiKey:

POAVRNPUPPJLKZ-UHFFFAOYSA-N

Formula:

C15H14O

SMILES:

OC1c2ccccc2CCc2ccccc21

Mol. weight [g/mol]:

210.27

CAS:

1210-34-0

## Physical Properties

Property code	Value	Unit	Source
gf	204.91	kJ/mol	Joback Method
hf	17.76	kJ/mol	Joback Method
hfus	24.13	kJ/mol	Joback Method
hvap	71.45	kJ/mol	Joback Method
log10ws	-3.96		Crippen Method
logp	2.867		Crippen Method
mcvol	169.700	ml/mol	McGowan Method
pc	3072.75	kPa	Joback Method
tb	704.84	K	Joback Method
tc	936.55	K	Joback Method
tf	415.45	K	Joback Method
vc	0.635	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	533.41	J/molxK	936.55	Joback Method
cpg	523.52	J/molxK	897.93	Joback Method
cpg	512.96	J/molxK	859.31	Joback Method
cpg	501.64	J/molxK	820.69	Joback Method
cpg	489.46	J/molxK	782.08	Joback Method

cpg	476.33	J/molxK	743.46	Joback Method
cpg	462.14	J/molxK	704.84	Joback Method
dvisc	0.0019016	Paxs	415.45	Joback Method
dvisc	0.0001092	Paxs	704.84	Joback Method
dvisc	0.0001476	Paxs	656.61	Joback Method
dvisc	0.0002093	Paxs	608.38	Joback Method
dvisc	0.0003151	Paxs	560.14	Joback Method
dvisc	0.0005124	Paxs	511.91	Joback Method
dvisc	0.0009221	Paxs	463.68	Joback Method
hfust	19.00	kJ/mol	365.20	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	440.70	K	0.10	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1210340&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1210340&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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