

# Benzene, 1,2-dimethyl-4-(phenylmethyl)-

<b>Other names:</b>	Methane, phenyl-3,4-xylyl-
<b>Inchi:</b>	InChI=1S/C15H16/c1-12-8-9-15(10-13(12)2)11-14-6-4-3-5-7-14/h3-10H,11H2,1-2H3
<b>InchiKey:</b>	IWSSFULMLQTJFO-UHFFFAOYSA-N
<b>Formula:</b>	C15H16
<b>SMILES:</b>	<chem>Cc1ccc(Cc2ccccc2)cc1C</chem>
<b>Mol. weight [g/mol]:</b>	196.29
<b>CAS:</b>	13540-56-2

## Physical Properties

Property code	Value	Unit	Source
gf	280.98	kJ/mol	Joback Method
hf	97.19	kJ/mol	Joback Method
hfus	21.91	kJ/mol	Joback Method
hvap	54.86	kJ/mol	Joback Method
log10ws	-4.52		Crippen Method
logp	3.894		Crippen Method
mcvol	174.690	ml/mol	McGowan Method
pc	2436.25	kPa	Joback Method
tb	605.92	K	Joback Method
tc	842.62	K	Joback Method
tf	336.69	K	Joback Method
vc	0.659	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	421.04	J/molxK	605.92	Joback Method
cpg	497.83	J/molxK	803.17	Joback Method
cpg	484.68	J/molxK	763.72	Joback Method
cpg	470.49	J/molxK	724.27	Joback Method
cpg	455.19	J/molxK	684.82	Joback Method
cpg	438.73	J/molxK	645.37	Joback Method
cpg	510.01	J/molxK	842.62	Joback Method
dvisc	0.0001596	Paxs	605.92	Joback Method

dvisc	0.0001994	Paxs	561.05	Joback Method
dvisc	0.0002590	Paxs	516.18	Joback Method
dvisc	0.0003535	Paxs	471.31	Joback Method
dvisc	0.0005152	Paxs	426.43	Joback Method
dvisc	0.0008204	Paxs	381.56	Joback Method
dvisc	0.0014790	Paxs	336.69	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13540562&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13540562&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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