

# 9-Anthracenemethanol

<b>Other names:</b>	9-Anthrylcarbinol 9-Hydroxymethylantracene 9-Hydroxymethylantracene 9-Methylolantracene 9-Anthracene carbinol
<b>Inchi:</b>	InChI=1S/C15H12O/c16-10-15-13-7-3-1-5-11(13)9-12-6-2-4-8-14(12)15/h1-9,16H,10H2
<b>InchiKey:</b>	JCJNNHDZTLRSGN-UHFFFAOYSA-N
<b>Formula:</b>	C15H12O
<b>SMILES:</b>	OCc1c2ccccc2cc2ccccc12
<b>Mol. weight [g/mol]:</b>	208.26
<b>CAS:</b>	1468-95-7

## Physical Properties

Property code	Value	Unit	Source
gf	245.05	kJ/mol	Joback Method
hf	90.57	kJ/mol	Joback Method
hfus	26.00	kJ/mol	Joback Method
hvap	72.54	kJ/mol	Joback Method
log10ws	-5.22		Crippen Method
logp	3.485		Crippen Method
mcvol	165.400	ml/mol	McGowan Method
pc	3173.97	kPa	Joback Method
tb	709.38	K	Joback Method
tc	936.88	K	Joback Method
tf	436.49	K	Joback Method
vc	0.630	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	431.07	J/molxK	709.38	Joback Method
cpg	442.90	J/molxK	747.30	Joback Method
cpg	453.92	J/molxK	785.21	Joback Method
cpg	464.21	J/molxK	823.13	Joback Method

cpg	473.88	J/molxK	861.05	Joback Method
cpg	483.03	J/molxK	898.96	Joback Method
cpg	491.77	J/molxK	936.88	Joback Method
dvisc	0.0014850	Paxs	436.49	Joback Method
dvisc	0.0008145	Paxs	481.97	Joback Method
dvisc	0.0004955	Paxs	527.45	Joback Method
dvisc	0.0003262	Paxs	572.93	Joback Method
dvisc	0.0002283	Paxs	618.42	Joback Method
dvisc	0.0001678	Paxs	663.90	Joback Method
dvisc	0.0001283	Paxs	709.38	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1468957&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1468957&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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