

# 2-ethyl-2-adamantanol

<b>Inchi:</b>	InChI=1S/C12H20O/c1-2-12(13)10-4-8-3-9(6-10)7-11(12)5-8/h8-11,13H,2-7H2,1H3
<b>InchiKey:</b>	YUBKBLFRGDNIDR-UHFFFAOYSA-N
<b>Formula:</b>	C12H20O
<b>SMILES:</b>	CCC1(O)C2CC3CC(C2)CC1C3
<b>Mol. weight [g/mol]:</b>	180.29

## Physical Properties

Property code	Value	Unit	Source
gf	62.58	kJ/mol	Joback Method
hf	-256.44	kJ/mol	Joback Method
hfus	19.07	kJ/mol	Joback Method
hvap	57.13	kJ/mol	Joback Method
log10ws	-2.94		Crippen Method
logp	2.584		Crippen Method
mcvol	153.230	ml/mol	McGowan Method
pc	2817.33	kPa	Joback Method
rinpol	1478.00		NIST Webbook
rinpol	1464.00		NIST Webbook
rinpol	1496.00		NIST Webbook
rinpol	1446.00		NIST Webbook
ripol	1953.00		NIST Webbook
ripol	1996.00		NIST Webbook
ripol	1974.00		NIST Webbook
tb	581.53	K	Joback Method
tc	783.10	K	Joback Method
tf	351.54	K	Joback Method
vc	0.586	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	439.53	J/mol×K	581.53	Joback Method
cpg	457.35	J/mol×K	615.12	Joback Method
cpg	474.05	J/mol×K	648.72	Joback Method

cpg	489.79	J/mol×K	682.31	Joback Method
cpg	504.70	J/mol×K	715.91	Joback Method
cpg	518.95	J/mol×K	749.50	Joback Method
cpg	532.67	J/mol×K	783.10	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R304546&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R304546&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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