

# 5,6-Dihydro-1,4-dimethylazulene

<b>Inchi:</b>	InChI=1S/C12H14/c1-9-5-3-4-6-11-10(2)7-8-12(9)11/h4,6-8H,3,5H2,1-2H3
<b>InchiKey:</b>	IQELVECEDPOJSE-UHFFFAOYSA-N
<b>Formula:</b>	C12H14
<b>SMILES:</b>	CC1=C2C=CCCC(C)=C2C=C1
<b>Mol. weight [g/mol]:</b>	158.24

## Physical Properties

Property code	Value	Unit	Source
gf	220.00	kJ/mol	Joback Method
hf	55.87	kJ/mol	Joback Method
hfus	15.90	kJ/mol	Joback Method
hvap	47.25	kJ/mol	Joback Method
log10ws	-4.05		Crippen Method
logp	3.539		Crippen Method
mvol	141.020	ml/mol	McGowan Method
pc	2890.51	kPa	Joback Method
rinpol	1425.00		NIST Webbook
rinpol	1425.00		NIST Webbook
tb	530.42	K	Joback Method
tc	759.63	K	Joback Method
tf	308.40	K	Joback Method
vc	0.535	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	314.36	J/mol×K	530.42	Joback Method
cpg	385.17	J/mol×K	721.43	Joback Method
cpg	372.87	J/mol×K	683.22	Joback Method
cpg	359.70	J/mol×K	645.02	Joback Method
cpg	345.60	J/mol×K	606.82	Joback Method
cpg	330.51	J/mol×K	568.62	Joback Method
cpg	396.66	J/mol×K	759.63	Joback Method
dvisc	0.0002954	Paxs	530.42	Joback Method

dvisc	0.0003485	Paxs	493.42	Joback Method
dvisc	0.0004224	Paxs	456.41	Joback Method
dvisc	0.0005295	Paxs	419.41	Joback Method
dvisc	0.0006935	Paxs	382.41	Joback Method
dvisc	0.0009624	Paxs	345.40	Joback Method
dvisc	0.0014447	Paxs	308.40	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R426196&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R426196&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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