

# 1,4-Methanonaphthalen-9-ol,1,2,3,4-tetrahydro-6-

<b>Inchi:</b>	InChI=1S/C12H14O/c1-7-2-3-8-9-4-5-10(12(9)13)11(8)6-7/h2-3,6,9-10,12-13H,4-5H2,1H
<b>InchiKey:</b>	ANPDCXWZMJATJ-UVTZGIPTSA-N
<b>Formula:</b>	C12H14O
<b>SMILES:</b>	Cc1ccc2c(c1)C1CCC2C1O
<b>Mol. weight [g/mol]:</b>	174.24
<b>CAS:</b>	1201-10-1

## Physical Properties

Property code	Value	Unit	Source
gf	132.38	kJ/mol	Joback Method
hf	-98.23	kJ/mol	Joback Method
hfus	23.63	kJ/mol	Joback Method
hvap	61.93	kJ/mol	Joback Method
ie	8.31 ± 0.02	eV	NIST Webbook
log10ws	-3.06		Crippen Method
logp	2.331		Crippen Method
mcvol	140.330	ml/mol	McGowan Method
pc	3173.97	kPa	Joback Method
tb	607.32	K	Joback Method
tc	816.51	K	Joback Method
tf	372.44	K	Joback Method
vc	0.539	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	377.41	J/molxK	607.32	Joback Method
cpg	439.17	J/molxK	781.65	Joback Method
cpg	428.32	J/molxK	746.78	Joback Method
cpg	416.81	J/molxK	711.92	Joback Method
cpg	404.53	J/molxK	677.05	Joback Method
cpg	391.43	J/molxK	642.19	Joback Method
cpg	449.42	J/molxK	816.51	Joback Method
dvisc	0.0006118	Paxs	607.32	Joback Method

dvisc	0.0007334	Paxs	568.17	Joback Method
dvisc	0.0009031	Paxs	529.03	Joback Method
dvisc	0.0011496	Paxs	489.88	Joback Method
dvisc	0.0015262	Paxs	450.73	Joback Method
dvisc	0.0021383	Paxs	411.59	Joback Method
dvisc	0.0032160	Paxs	372.44	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1201101&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1201101&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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