

# «alpha»-Methyl-2-naphthalenemethanol

<b>Other names:</b>	1-(2-Naphthyl)ethanol Methyl 2-naphtylcarbinol «beta»-Naphthyl methyl carbinol 2-Naphthalenemethanol, «alpha»-methyl- 1-(Naphthalen-2-yl)ethanol «alpha»-methylnaphthalene-2-methanol
<b>Inchi:</b>	InChI=1S/C12H12O/c1-9(13)11-7-6-10-4-2-3-5-12(10)8-11/h2-9,13H,1H3
<b>InchiKey:</b>	AXRKCRWZRKETCK-UHFFFAOYSA-N
<b>Formula:</b>	C12H12O
<b>SMILES:</b>	CC(O)c1ccc2ccccc2c1
<b>Mol. weight [g/mol]:</b>	172.22
<b>CAS:</b>	7228-47-9

## Physical Properties

Property code	Value	Unit	Source
gf	120.33	kJ/mol	Joback Method
hf	-32.39	kJ/mol	Joback Method
hfus	18.07	kJ/mol	Joback Method
hvap	63.17	kJ/mol	Joback Method
log10ws	-3.80		Crippen Method
logp	2.893		Crippen Method
mcvol	142.590	ml/mol	McGowan Method
pc	3423.86	kPa	Joback Method
tb	616.34	K	Joback Method
tc	833.18	K	Joback Method
tf	342.46	K	Joback Method
vc	0.534	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	348.62	J/mol×K	616.34	Joback Method
cpg	360.97	J/mol×K	652.48	Joback Method
cpg	372.47	J/mol×K	688.62	Joback Method

cpg	383.17	J/molxK	724.76	Joback Method
cpg	393.14	J/molxK	760.90	Joback Method
cpg	402.45	J/molxK	797.04	Joback Method
cpg	411.15	J/molxK	833.18	Joback Method
dvisc	0.0045007	Paxs	342.46	Joback Method
dvisc	0.0016725	Paxs	388.11	Joback Method
dvisc	0.0007655	Paxs	433.75	Joback Method
dvisc	0.0004066	Paxs	479.40	Joback Method
dvisc	0.0002411	Paxs	525.05	Joback Method
dvisc	0.0001554	Paxs	570.69	Joback Method
dvisc	0.0001069	Paxs	616.34	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7228479&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7228479&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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