

2,6-Difluorobenzoic acid, 3-methylbut-2-enyl ester

Inchi:	InChI=1S/C12H12F2O2/c1-8(2)6-7-16-12(15)11-9(13)4-3-5-10(11)14/h3-6H,7H2,1-2H3
InchiKey:	YQNYJIWZPQVFM-T-UHFFFAOYSA-N
Formula:	C12H12F2O2
SMILES:	CC(C)=CCOC(=O)c1c(F)cccc1F
Mol. weight [g/mol]:	226.22

Physical Properties

Property code	Value	Unit	Source
gf	-408.56	kJ/mol	Joback Method
hf	-607.01	kJ/mol	Joback Method
hfus	27.94	kJ/mol	Joback Method
hvap	53.47	kJ/mol	Joback Method
log10ws	-3.91		Crippen Method
logp	3.088		Crippen Method
mcvol	162.860	ml/mol	McGowan Method
pc	2365.67	kPa	Joback Method
rinpol	1481.30		NIST Webbook
tb	589.47	K	Joback Method
tc	790.29	K	Joback Method
tf	330.76	K	Joback Method
vc	0.640	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	389.54	J/mol×K	589.47	Joback Method
cpg	402.53	J/mol×K	622.94	Joback Method
cpg	414.79	J/mol×K	656.41	Joback Method
cpg	426.36	J/mol×K	689.88	Joback Method
cpg	437.26	J/mol×K	723.35	Joback Method
cpg	447.51	J/mol×K	756.82	Joback Method
cpg	457.13	J/mol×K	790.29	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U292582&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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