

# Benzene, 1-methyl-4-(1-chloro-1-methylethyl)

<b>Inchi:</b>	InChI=1S/C10H13Cl/c1-8-4-6-9(7-5-8)10(2,3)11/h4-7H,1-3H3
<b>InchiKey:</b>	ITTZHNHXCRWVKL-UHFFFAOYSA-N
<b>Formula:</b>	C10H13Cl
<b>SMILES:</b>	Cc1ccc(C(C)(C)Cl)cc1
<b>Mol. weight [g/mol]:</b>	168.66

## Physical Properties

Property code	Value	Unit	Source
gf	127.01	kJ/mol	Joback Method
hf	-49.16	kJ/mol	Joback Method
hfus	12.09	kJ/mol	Joback Method
hvap	43.88	kJ/mol	Joback Method
log10ws	-3.49		Crippen Method
logp	3.469		Crippen Method
mcvol	140.240	ml/mol	McGowan Method
pc	2829.33	kPa	Joback Method
tb	494.06	K	Joback Method
tc	720.15	K	Joback Method
tf	273.74	K	Joback Method
vc	0.525	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	288.69	J/molxK	494.06	Joback Method
cpg	303.90	J/molxK	531.74	Joback Method
cpg	318.04	J/molxK	569.42	Joback Method
cpg	331.18	J/molxK	607.10	Joback Method
cpg	343.37	J/molxK	644.79	Joback Method
cpg	354.69	J/molxK	682.47	Joback Method
cpg	365.18	J/molxK	720.15	Joback Method
dvisc	0.0033165	Paxs	273.74	Joback Method
dvisc	0.0016285	Paxs	310.46	Joback Method
dvisc	0.0009295	Paxs	347.18	Joback Method

dvisc	0.0005906	Paxs	383.90	Joback Method
dvisc	0.0004062	Paxs	420.62	Joback Method
dvisc	0.0002967	Paxs	457.34	Joback Method
dvisc	0.0002270	Paxs	494.06	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R131878&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R131878&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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