

# 1,4-Epoxy-naphthalene, 1,2,3,4-tetrahydro-

<b>Other names:</b>	1,4-Epoxy-1,2,3,4-tetrahydronaphthalene 1,2,3,4-Tetrahydronaphthalene 1,4-endoxide 1,2,3,4-Tetrahydronaphthalene-1,4-endo-oxide
<b>Inchi:</b>	InChI=1S/C10H10O/c1-2-4-8-7(3-1)9-5-6-10(8)11-9/h1-4,9-10H,5-6H2
<b>InchiKey:</b>	SPELXJYTVLDKGB-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O
<b>SMILES:</b>	c1ccc2c(c1)C1CCC2O1
<b>Mol. weight [g/mol]:</b>	146.19
<b>CAS:</b>	35185-96-7

## Physical Properties

Property code	Value	Unit	Source
gf	183.58	kJ/mol	Joback Method
hf	-4.91	kJ/mol	Joback Method
hfus	21.66	kJ/mol	Joback Method
hvap	44.96	kJ/mol	Joback Method
log10ws	-2.87		Crippen Method
logp	2.593		Crippen Method
mcvol	112.150	ml/mol	McGowan Method
pc	3736.23	kPa	Joback Method
tb	496.02	K	Joback Method
tc	727.10	K	Joback Method
tf	307.37	K	Joback Method
vc	0.430	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	258.11	J/mol×K	496.02	Joback Method
cpg	273.47	J/mol×K	534.53	Joback Method
cpg	287.55	J/mol×K	573.05	Joback Method
cpg	300.47	J/mol×K	611.56	Joback Method
cpg	312.33	J/mol×K	650.07	Joback Method
cpg	323.25	J/mol×K	688.59	Joback Method

cpg	333.35	J/mol×K	727.10	Joback Method
dvisc	0.0014369	Paxs	307.37	Joback Method
dvisc	0.0013461	Paxs	338.81	Joback Method
dvisc	0.0012752	Paxs	370.25	Joback Method
dvisc	0.0012183	Paxs	401.69	Joback Method
dvisc	0.0011716	Paxs	433.14	Joback Method
dvisc	0.0011327	Paxs	464.58	Joback Method
dvisc	0.0010998	Paxs	496.02	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	327.20	K	0.50	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C35185967&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C35185967&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

**tbrp:** Boiling point at reduced pressure  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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